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(54) BENZAMIDOXIME DERIVATIVE, PROCESS FOR PRODUCTION THEREOF, AND AGROHORTICULTURAL BACTERICIDE

BENZAMIDOXIMDERIVATE, VERFAHREN ZU IHRER HERSTELLUNG UND EIN BAKTERIZID FÜR LANDWIRTSCHAFT UND GARTENBAU

DERIVE DE BENZAMIDOXIME, PROCEDE DE PRODUCTION, ET BACTERICIDE AGROHORTICOLE

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(56) References cited: JP-A- 2 006 453

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Description

Technical Field:

[0001] The present invention relates to novel benzamidoxime derivatives, methods for preparation thereof and fungicides for agricultural and horticultural uses.

Background Art:

[0002] In farming of agricultural and horticultural crops in the past, various fungicides have been used for the control of plant diseases on the crops, however, many of them are not enough useful because of their insufficient effectiveness in plant disease control, the limitation in their use due to the appearance of resistant strain of plant disease pathogens to the fungicides, the development of phytotoxicity and contamination to the crops, and/or their strong toxicity to humans, domestic animals and wildlife. For this reason, there is still intensive requirement to develop safe fungicides for agricultural and horticultural uses, which do not have the disadvantages as described above.

[0003] Some benzamidoxime derivatives, which are close to the compounds of the present invention, and their use as fungicides have been disclosed in Japanese Patent Laid-opened No. Hei 2-6453 Gazette. However, it is obvious that the biological activity of those benzamide oxime derivatives are not enough in the practical plant disease control. [0004] Therefore, it is an object of the present invention to provide novel compounds which can be a fungicide for agricultural and horticultural use capable of advantageously manufacturing the compound in an industrial scale, controlling plant diseases steadily and using it safely.

Disclosure of the Invention:

[0005] The present invention is directed to benzamidoxime derivatives represented by a general formula [i];

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wherein R¹ is unsubstituted or substituted $C_1 - C_4$ alkyl, unsubstituted or substituted $C_2 - C_4$ alkenyl or unsubstituted or substituted $C_2 - C_4$ alkynyl, R² is phenyl optionally having substituents or heterocycle optionally having substituents, X¹is-CF₃ X², X³, X⁴ and X⁵ are each independently hydrogen, halogen, $C_1 - C_4$ alkyl, $C_1 - C_4$ haloalkyl, $C_1 - C_4$ alkylsulfinyl, $C_1 - C_4$ alkylsulfinyl, nitro, amino or $C_1 - C_4$ alkylcarbonylamino, and r¹ and r² are each independently hydrogen, halogen, $C_1 - C_4$ alkyl, $C_1 - C_4$ haloalkyl, $C_1 - C_4$ alkoxy, $C_1 - C_4$ alkylthio or amino, or r¹ and r² together may form a carbonyl.

[0006] The present invention is also directed to methods for preparation thereof and fungicides for agricultural and horticultural use comprising the said derivatives.

[0007] In the present invention, for the examples of C₁ - C₄ alkyl of the unsubstituted or substituted C₁ - C₄ alkyl represented by R¹, methyl, ethyl, propyl, isopropyl, butyl, isobutyl and t-butyl can be given.

[0008] For the examples of C_2 - C_4 alkenyl of the unsubstituted or substituted C_2 - C_4 alkenyl represented by R^1 , vinyl, 1-propenyl, 2-propenyl, isopropenyl, 1-butenyl, 2-butenyl and 3-butenyl can be given.

[0009] For the examples of C_2 - C_4 alkynyl of the unsubstituted or substituted C_2 - C_4 alkynyl represented by R^1 , ethynyl, propargyl, 2-butynyl and 3-butynyl can be given.

[0010] Further, according to formula I' R^1 may be a group represented by general formula R^3 - CH_2 wherein R^3 is a group selected from; C_3 - C_8 cycloalkyl, such as cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl; C_3 - C_8 halocycloalkyl, such as 1-fluorocyclopropyl, 2-fluorocyclopropyl, 1-chlorocyclopropyl, 2-chlorocyclopropyl, 2,2-difluorocyclopropyl, 2-fluorocyclopentyl, 3-fluorocyclopentyl, 2-chlorocyclopentyl, 3-chlorocyclopentyl, 3-difluorocyclohexyl, 3,4-difluorocyclohexyl, 3,4-dichlorocyclohexyl and 3,4-dibromosilocyclohexy 1; C_3 - C_8 cycloalkenyl, such as 2-cyclohexenyl and 3-cyclohexenyl; halogens, such as fluorine, chlorine, bromine and lodine; C_1 - C_4 alkoxy, such as methoxy, ethoxy, propyloxy, isopropyloxy, butyloxy, isobutyloxy and t-butyloxy; unsubstituted, mono-substituted or di-substituted amino by C_1 - C_4 alkyl, such as amino, methylamino and dimethylamino; unsubstituted, mono-substituted

or di-substituted carbamoyl by C_1 - C_4 alkyl, such as carbamoyl, methylcarbamoyl and dimethylcarbamoyl; C_1 - C_4 alkylthio, such as methylthio, ethylthio, propylthio and isopropylthio; C_1 - C_4 alkylsulfinyl, such as methylsulfinyl and ethylsulfinyl; C_1 - C_4 alkylsulfonyl, such as methylsulfonyl and ethylsulfonyl; C_1 - C_4 alkoxycarbonyl, such as methoxycarbonyl and ethoxycarbonyl; carboxy and cyano, can be given for the examples of the substituents for any of C_1 - C_4 alkyl, C_2 - C_4 alkeyl and C_2 - C_4 alkynyl.

[0011] However, straight chain or branched C_1 - C_4 alkyl being unsubstituted or substituted is more preferable for the examples of the substituent represented by R^1 . More particularly, straight chain or branched C_1 - C_4 alkyl, such as methyl, ethyl, propyl, isopropyl, butyl, s-butyl and t-butyl; a group represented by a general formula, R^3CH_2 , wherein R^3 is C_3 - C_8 cycloalkyl, C_1 - C_3 haloalkyl, C_1 - C_3 alkoxy, C_1 - C_3 alkylthio, C_1 - C_3 alkylsulfinyl, C_1 - C_3 alkylsulfinyl, C_1 - C_3 alkylsulfinyl, cycloalkylmethyl including cyclopropylmethyl, cyclobutylmethyl, cyclopentylmethyl and cyclohexylmethyl, C_1 - C_8 cycloalkylmethyl including 2-fluorocyclopropylmethyl, 1-fluorocyclopropylmethyl, 1,2-difluorocyclopropylmethyl and 3,4-dibromocyclohexyl, C_1 - C_4 haloalkyl including 2-chloroethyl, 2-fluoroethyl, 2,2-dichloroethyl, 2,2-difluoroethyl and 2,2,2-trifluoroethyl, C_1 - C_4 alkoxymethyl including methoxymethyl, ethoxymethyl and propoxymethyl, C_2 - C_4 alkynyl including propargyl, C_2 - C_4 alkenyl including allyl, 2-butenyl, cyanomethyl, alkoxycarbonylmethyl including methoxycarbonylmethyl and ethylsulfinylmethyl and ethylsulfinylmethyl and ethylsulfinylmethyl including methylsulfinylmethyl including methylsulfonylmethyl and ethylsulfonylmethyl, alkylsulfinylmethyl, alkylsulfinylmethyl, aminomethyl, substituted aminomethyl, lincluding N-methylaminomethyl, N,N-dimethylaminomethyl, N-acetylaminomethyl and N-benzoylaminomethyl, can be given for the examples of the straight chain or branched C_1 - C_4 alkyl being unsubstituted or unsubstituted described above.

[0012] For the examples of heterocycle of the unsubstituted or substituted heterocycle group represented by R², 5-or 6-membered aromatic heterocycle containing 1 - 4 heteroatoms, such as N. 0 and S, such as pyridine ring, fran ring, thiophene ring, pyrazole ring, imidazole ring, triazole ring, pyrrole ring, pyrazine ring, pyrimidine ring, pyridazine ring, oxazole ring, isoxazole ring and thiazole ring, can be given.

[0013] The substituents for phenyl and heterocycle represented by R^2 may substitute one or more optional positions of the benzene ring or the heterocycle thereof and may be different with each other if 2 or more positions are substituted thereby. For the preferable examples of the substituents described hereinabove, halogens, such as fluorine, chlorine and bromine, $C_1 - C_4$ alkyl, such as methyl, ethyl, propyl, isopropyl, butyl and t-butyl, $C_1 - C_4$ alkoxy, such as methoxy, ethoxy, propoxy, isopropoxy, butoxy and t-butoxy, $C_2 - C_4$ alkenyloxy, such as allyoxy and crotyloxy, $C_2 - C_4$ alkynyloxy, such as propargyloxy, $C_1 - C_4$ haloalkyl, such as chloromethyl, fluoromethyl, bromomethyl, dichloromethyl, difluoromethyl, trichloromethyl, trifluoromethoxy, and $C_1 - C_4$ haloalkoxy, such as chloromethoxy, fluoromethoxy, bromomethoxy, dichloromethoxy, difluoromethoxy, trichloromethoxy, trifluoromethoxy, trifluoromethoxy, trifluoromethoxy, trifluoromethoxy, trifluoromethoxy, trifluoromethoxy, tribromomethoxy and trifluoroethoxy, can be given.

[0014] For the examples of halogen atoms represented by X^2 , X^3 , X^4 and X^5 , fluorine, chlorine, bromine and iodine can be given, and for the examples of the C_1 - C_4 alkyl represented by X^2 , X^3 , X^4 and X^5 , methyl, ethyl, propyl, isopropyl, butyl, isobutyl and t-butyl can be given, and further for the examples of the C_1 - C_4 haloalkyl represented by X^2 , X^3 , X^4 and X^5 , straight chain or branched C_1 - C_4 haloalkyl, such as chloromethyl, dichloromethyl, trichloromethyl, difluoromethyl, trifluoromethyl, bromomethyl, dibromomethyl, chloroethyl, fluoroethyl, dichloroethyl, difluoroethyl, tetrafluoroethyl, pentafluoroethyl, chloropopyl, perfluoropopyl, perfluorosopropyl, chlorobutyl, fluorobutyl, perfluorobutyl, perfluorosobutyl, fluorosobutyl, perfluoro-s-butyl, fluoro-s-butyl, perfluoro-s-butyl, chloro-t-butyl and perfluoro-t-butyl, can be given, and still for the examples of the C_1 - C_4 alkoxy represented by X^2 , X^3 , X^4 and X^5 , methoxy, ethoxy, propyloxy, isopropyloxy, butyloxy, isobutyloxy and t-butyloxy can be given. In addition, for the examples of the C_1 - C_4 alkylthio represented by X^2 , X^3 , X^4 and X^5 , methylthio, ethylthio, propylthio, isopropylthio, butylthio, isobutylthio and t-butylthio can be given, and for the examples of the C_1 - C_4 haloalkoxy represented by X^2 , X^3 , X^4 and X^5 , trifluoromethoxy, difluoromethoxy, trichloromethoxy, trifluoroethoxy and tetrafluoroethoxy can be given.

[0015] Furthermore, for the examples of the groups represented by r^1 and r^2 , which may be the same or different groups with each other, hydrogen, halogen atoms, such as fluorine, chlorine, bromine and iodine, $C_1 - C_4$ alkyl, such as methyl, ethyl, propyl, isopropyl, butyl and t-butyl, $C_1 - C_4$ alkoxy, such as methoxy, ethoxy, propoxy, isopropoxy, butoxy and t-butoxy, $C_1 - C_4$ alkylthio, such as methylthio, ethylthio, propylthio, ispropylthio, butylthio and t-butylthio, $C_1 - C_4$ haloalkyl, such as trifluoromethyl, trichloromethyl, tribromomethyl, trifluoroethyl, chloromethyl, fluoromethyl and pentafluoroethyl, and amino can be given.

[0016] Moreover, r1 and r2 may in together form a carbonyl group.

(Manufacturing of the Compounds)

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[0017] The compounds of the present invention can be manufactured according to the following reaction formula;

wherein Hal represents halogen, and R1, R2, X2, X3, X4, X5, r1 and r2 are as described above.

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[0018] The reaction described above is carried out by reacting the compound represented by the general formula [II] and the compound represented by the general formula [III] in an organic solvent for from 10 min. to several dozens of hours at a temperature range of from 0°C to the boiling point of the solvent used and in the presence of a base, if appropriate.

[0019] For the examples of the solvent usable in the reaction described above, aromatic hydrocarbons, such as benzene and toluene, ethers, such as THF and diethyl ether, halogenated hydrocarbons, such as chloroform and dichloro methane, amides, such as DMF, DMSO, acetonitrile and the mixture of the solvent exemplified above, can be given.

[0020] As the base usable in the reaction, pyridine, triethyl amine, DBU, sodium hydroxide, sodium carbonate, potassium carbonate or the like can be exemplified.

[0021] After completed the reaction, the objective compounds can be obtained by taking an ordinary post-treatment procedure and allowing the products to the purification by using silica gel column chromatography or the else.

[0022] Whereas, the raw material compound of the present invention, represented by the general formula [II] can be synthesized according to the following reaction formula;

wherein L is an eliminating group, such as paratoluenesulfonyloxy, methylsulfonyloxy and halogen atoms, and R^1 , X^2 , X^3 , X^4 and X^5 are as described above.

[0023] The first step reaction in the reaction formula shown above is to obtain a benzamide oxime compound represented by the general formula [V], wherein a nitrile compound represented by the general formula [IV] and hydroxyl amine hydrochloride are allowed to a reaction for 10 min. to several dozens of hours in an inactive solvent and in the presence of a base at a temperature range of from 0 °C to the boiling point of the solvent used.

[0024] For the examples of the solvent usable in the reaction described above, alcohols, such as methanol, ethanol and propanol, ethers, such as THF and diethyl ether, amides, such as DMF, DMSO, water and the mixture of the solvent exemplified hereinabove, can be given.

[0025] Further, for the examples of the base usable in the reaction described above, sodium carbonate, sodium

hydrogencarbonate, potassium carbonate, sodium hydroxide, potassium hydroxide triethyl amine, pyridine and the like can be given.

[0026] The second step reaction in the reaction formula shown above is to obtain the raw material compound represented by the general formula [II]. wherein the compound represented by the general formula [V] and a compound represented by a general formula, R¹-L, are allowed to a reaction for 10 min. to several dozens of hours in a solvent and in the presence of a base at a temperature range of from -15 °C to the boiling point of the solvent used.

[0027] For the examples of the base usable in the second step reaction, metal alkoxide, such as sodium methoxide and sodium ethoxide. inorganic bases, such as sodium hydride, sodium hydroxide, potassium hydroxide and potassium carbonate, and organic bases, such as triethyl amine and pyridine, can be given.

[0028] Furthermore, if appropriate, catalysts may be used in the second step reaction, though it depends on the type of the solvent and the base to be used. For the examples of the catalysts usable in the reaction, crown ethers, such as 18-crown-6 and dicyclohexyl-18-crown-6, tetrabutyl ammonium bromide and other chlorides, quaternary ammonium salts, such as methyltrioctyl ammonium chloride and benzyltriethyl ammonium chloride, and phosphonium salts, such as tetraphenyl phosphonium bromide and hexadecyltributyl phosphonium iodide, can be given.

[0029] The chemical structures of the compounds of the present invention were determined by using NMR, IR, MS and the other analytical apparatuses.

Best Mode for Carrying Out the Invention:

[0030] Now, the present invention is further described in detail with referring to the examples described hereinbelow.

Example 1

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Preparation of N'-cyclopropylmethyloxy-N-(4-methoxyphenyl)acetyl-2-fluoro-6-trifluoromethylbenzamidine (Compound No. 56)

[0031]

CF₃ NOCH₂ + CH₃O -

[0032] In 200ml of benzene was dissolved 20.0g of N'-cyclopropylmethyloxy-2-fluoro-6-trifluoromethylbenzamidine, and to the solution was added 16. 0g of 4-methoxyphenylacetylchloride. The solution was heated under refluxing for 10 hours. After cooling, ethyl acetate was added to the solution, followed by washing with water and drying over anhydrous magnesium sulfate. The organic layer was concentrated under reduced pressure and the residue was subjected to silica gel column chromatography to obtain 23.6g of the objective compound.m.p. 75-76°C

Example 2

Preparation N'-cyclopropylmethyloxy-N-(4-methoxyphenyl)acetyl-2-trifluoromethyl benzamidine (Compound No. 12)

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[0034] In 80ml of benzene was dissolved 10.4g of N'-cyclopropylmethyloxy-2-fluoro-6-trifluoromethylbenzamidine, and to the solution was added 8.9g of 4-methoxyphenylacetylchloride. The solution was heated under refluxing for 3 hours. After cooling, ethyl acetate was added to the solution, followed by washing with water and drying over anhydrous magnesium sulfate. The organic layer was concentrated under reduced pressure and the residue obtained in crystal was washed with a mixed solvent hexane and ether to thereby obtain 13.7g of crude crystal. The crystal was then recrystallized in hexane, thereby affording 11.5g of the objective compound.m.p. 88-90 °C

[0035] Now, the examples for preparing the raw material compounds described above to be used for the preparing of the compounds of the present invention are described hereinbelow.

Reference Example 1

Preparation of 2-fluoro-6-trifluoromethylbenzamidoxime

[0036]

[0037] In 540ml of methanol was dissolved 58.8g of hydroxylamine hydrochloride and to the solution was added 160ml of aqueous solution of sodium carbonate containing 49.4g thereof. 40g of 2-fluoro-6-trifluoromethylbenzonitrile was added thereto at room temperature with stirring, and then further stirred 3 hours at 60°C. After removing the solvent methanol by distillation from the solution, the solution was extracted with ethyl acetate. The organic layer was dried over anhydrous magnesium sulfate, and it was concentrated under reduced pressure, thereby obtaining crude crystals. The crystals were then added with 200ml of 3N aqueous solution of hydrochloric acid and thoroughly stirred, then the insoluble substance resulted in the solution was removed by filtration. The filtrate was then neutralized with 10% aqueous solution of sodium hydroxide under cooling and then extracted again with ethyl acetate. The organic layer was dried over anhydrous magnesium sulfate, and the solution was concentrated under reduced pressure to obtain 26.6g of the objective compound.

m.p. 155-157°C

[0038] The examples of benzamide oxime derivatives represented by the general formula [V], which can be manu-

factured according to the methods as described above, are given in Table 1 hereinbelow.

Table 1.

Chemical	formula	X3 CF3	NOH NH2	
X²	X 3	X4	X 5	Physical const.
Н	H	Н	H	mp 124-126℃
H ·	H	H	Cl	mp 112-115℃
н	H	C1	F	mp 107~108℃
Н	H	CI	CI	n b 1.5210
H	H	H	F	mp 155~157℃
H	H	F	P	mp 105~107℃
H	H	F	Cl	mp 98~ 99°C
Н	Н	CF3	C1	mp 97- 99℃

Reference Example 2

Preparation of N'-cyclopropylmethyloxy-2-fluoro-6-trifluoromethylbenzami dine

[0039]

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[0040] In 100ml of DMF was dissolved 26.6g of 2-fluoro-6-trifluoromethyl-benzamidoxime and 17.8g of cyclopropyl-methylbromide, and to the solution was added 4.8g of sodium hydride (60% in oil) at 10 °C over 30 min. The solution was then stirred for 3 hours, and the solution reacted was poured into ice-water and extracted with ethyl acetate. The organic layer was washed with water and dried over anhydrous magnesium sulfate. The organic layer was concentrated under reduced pressure and the residue was subjected to silica gel column chromatography to obtain 24.8g of the objective compound.

m.p. 63-64°C

Reference Examples 3

Preparation of N'-cyclopropylmethyloxy-3,6-bistrifluoromethyl-2-chloro-benzamidine

5 Preparation of N'-cyclopropylmethyloxy-3,6-bistrifluoromethyl-2-chloro benzamidine

[0041]

according to the methods similar to the methods as described above, are given in Table 2 hereinbelow.

[0042] In 10ml of chloroform was dissolved 0.60g of 3,6-bistriftuoromethyl-2-chlorobenzamidoxime and 0.50g of cyclopropylmethylbromide, and to the solution was added 0.1g of tetrabutylammonium bromide at room temperature with stirring, and then 1.2ml of 10% aqueous solution of sodiumhydroxide, then stirred for 3 hours at 30-40°C. The solution was washed with water, washed with saturated saline solution and dried over anhydrous magnesium sulfate. The organic layer was concentrated under reduced pressure and the residue was subjected to silica gel column chromatography to obtain 0.40g of the objective compound.m.p. 75-80°C [0043] The examples of benzamidoxime derivatives represented by the general formula [II], which can be prepared

Table 2.

Chemi	ical fo	rmula	X ² CF ₃	NOR' NH2	
X²	X 3	Χ·	X ⁵	R¹	Physical const.
H	Н	H	H	CH ₂ cPr	n _p 1.4917
H	н	R	F	Bt	mp 64-66℃
H	H	H	F	iPr	n b 1.4789
H	H	H	F	CH2C(CH3)3	mp 97-98℃
Н	H	Н	F	CH2CH=CH2	mp 69-70℃
H	H	H	P	$CH_2C = CH$	n p 1.5011
H	H	H	Cl	CH2cPr	mp 43-46°C
н	н	Cl	P	CH2cPr	mp 71-73℃
Н	H	Cl	Ci	CH₂C = CH	n 0 1.5360
Cl	Н	н	Cl	CH₂cPr	n _p 1.5308
H	H	C1	Cl	CH₂cPr	mp 73-75℃
H	H	F	F	CH2cPr	mp 43-45℃
Н	H	P	Cl	CH2cPr	mp 54-55℃
H	Н	CF;	C1	CH ₂ cPr	mp 75-78℃

^{*}cPr represnts cyclopropylmethyl in the tables.

[0044] Now, the representative examples for the compounds of the present invention, which can be manufactured according to the praparation methods similar to the ones described in Examples 1 and 2, are given in Tables 3 and 4. However, it should be noted that X^1 , X^2 , X^3 , X^4 , X^5 , R^1 , R^2 , r_1 and r_2 given in the Tables 3 and 4 correspond to X^1 , X^2 , X^3 , X^4 , X^5 , R^1 , R^2 , r_1 and r_2 given for the compounds represented by the general formula [I], respectively.

Table 3.

	(r ¹ , r ² =H)													
10	No.	X¹	X ²	Хз	X ⁴	χ5	R¹	R ²	Physical const. mp (°C) refractive					
			 	 	<u> </u>				index					
	1	CF ₃	Н	H	Н	Н	Et	Ph-4-OMe	72-73					
	2	CF ₃	H	H	Н	Н	Me	Ph-4-OMe						
15	3	CF ₃	H	H	Н	H	CH ₂ cPr	2-thienyl	93-95					
	4	CF ₃	H	H	H	Н	CH ₂ cPr	3-thienyl	92-93					
	5	CF ₃	H	H	Н	H	CH ₂ cPr	Ph-2,4-F ₂	100-101					
	6	CF ₃	H	H	Н	H	CH ₂ cPr	Ph-2-F	84-85					
	7	CF ₃	H	H	H	Н	CH ₂ cPr	Ph-2-F,4-OMe	84-85					
20	8	CF ₃	H	Н	H	Н	CH ₂ cPr	Ph-3-Me	81-82					
	9	CF ₃	H	H	H	H	CH ₂ cPr	Ph-3-Me-4-OMe	77-79					
	10	CF ₃	Н	H	H	н	CH ₂ cPr	Ph-4-F	113-114					
	11	CF ₃	H	H	Н	H	CH ₂ cPr	Ph-4-Me	80-81					
25	12	CF ₃	H	Н	Н	Н	CH ₂ cPr	Ph-4-OMe	88-90					
	13	CF ₃	Н	Н	Н	Н	CH ₂ cPr	Ph	100-102					
	14	CF ₃	Н	H	н	Н	nBu	Ph-4-OMe						
	15	CF ₃	Н	H	н	Н	tBu	Ph-4-OMe	27.0					
	16	CF ₃	Н	H	Н	F	CH ₂ CH=CH ₂	Ph-4-OMe	n ^{27.0} 1.5290					
30	17	CF ₃	Н	Н	Н	F	CH ₂ CH=CH ₂	Ph	n ^{28.5} 1.5132					
	18	CF ₃	Н	Н	Н	F	CH(CH ₃)CH=CH ₂	Ph-4-OMe	l 76-78 l					
	19	CF ₃	Н	H	н	F	CH ₂ CH=CHCI	Ph-4-OMe	n ^{25.4} 1.5333					
	20	CF ₃	н	Н	Н	F	CH ₂ CH=CCl ₂	Ph-4-OMe	n 1.5333 n ^{29.7} 1.5362					
	21	CF ₃	Н	Н	Н	F	CH ₂ CH=CH-CH ₃	Ph-4-OMe	n ^{25/2} 1. 5148					
35	22	CF ₃	н	Н	н	F	Et	Ph-4-OMe	70-73					
	23	CF ₃	Н	Н	н	F	Et	Ph	59-61					
	24	CF ₃	Н	Н	Н	F	CH ₂ CH ₂ CI	Ph-4-OMe	n ^{24.0} 1.5330					
İ	25	CF ₃	н	н	н	F	CH ₂ CHF ₂	Ph-4-OMe	78-80					
40	26	CF ₃	н	н	н	F	Me	Ph-4-OMe						
	27	CF ₃	н	Н	н	F	CH ₂ C(CI)=CH ₂	Ph-4-OMe	n ^{25.5} 1.5242					
	28	CF ₃	Н	Н	н	F	CH ₂ C(CH ₃)=CH ₂	Ph-4-OMe	n 1.5242 n ²⁵⁰ 1.5162					
	29	CF3	Н	н	Н	F	CH ₂ CN	Ph-4-OMe	n 1.5162 n 29.5 1.5113					
	30	CF ₃	н	н	Н	F	CH ₂ CN	Ph	_23:5 4 FOOC					
45	31	CF ₃	Н	Н	Н	F	CH ₂ OCH ₃	Ph-4-OMe	n ^{24.0} 1.5288					
	32	CF ₃	Н	н	Н	F	CH ₂ OCH ₃	Ph	_24.5 4 5070					
	33	CF ₃	н	н	н	F	CH ₂ cPr	3-methyl	n ^{24.8} 1.5133					
		•					_	pyrazol-1-yl	, , , , , , , , , , , , , , , , , , ,					
50	34	CF ₃	н	н	Н	F	CH ₂ cPr	4-methyl	n ^{23.6} 1.5121					
50							_	pyrazol-1-yl	l l					
	35	CF ₃	н	н	Н	F	CH ₂ cPr	pyrazol-1-yl	n ^{22.9} 1.5126					
	36	CF ₃	н	Н	н	F	CH ₂ cPr	3-methyl	n ^{29.2} 1.5310					
								2-thienyl						
55	37	CF ₃	Н	Н	н	F	CH ₂ cPr	4-methyl 2-thienyl	n ^{23.2} 1.5313					

Table 3. (continued)

	(r¹, r² = H)												
5	No.	X ¹	X ²	X ₃	X ⁴	X ⁵	R ¹	H ²	Physical const. mp (°C) refractive index				
	38	CF ₃	н	H	Н	F	CH ₂ cPr	5-methyl 2-thienyl	n ^{23.9} 1.5353				
	39	6 5	. H	н	н	F	CH alla	2-thienyl	n ^{22.2} 1.5346				
10	40	CF ₃	н	н	н	F	CH₂cPr CH₂cPr	4-methyl	n 1.5346 n ^{26.5} 1.5302				
	41	CF ₃	н	н	н	F	CH ₂ cPr	3-thienyl 5-methyl 3-thienyl	57-58				
15	42	CF ₃	Н	н	н	F	CH ₂ cPr	3-thienyi	70-72				
	43	CF ₃	Н	Н	Н	F	CH ₂ cPr	Ph-2,4- F ₂	n ^{26.0} 1 5083				
	44	CF ₃	Н	H	Н.	F	CH ₂ cPr	Ph-2-F	n ^{26:0} 1 5191				
	45	CF ₃	H	Н	н	F	CH ₂ cPr	Ph-2-F-3-Me	p.28.4 1 5107				
	46	CF ₃	Н	Н	н	F	CH ₂ cPr	Ph-2-F-4-OMe	29.5 4 5400				
20	47	CF ₃	Н	н	н	F	CH ₂ cPr	Ph-2-F-4-OMe 5-Me	n 1.5193 n ^{27.0} 1.6190				
	48	CF ₃	н	н	н	F	CH ₂ cPr	Ph-2-F-5-Me	n ^{26.7} 1.5143				
	49	CF ₃	;;	н	Н	F	CH ₂ cPr	Ph-3,5-Me ₂	88-89				
	50	CF ₃	l H	н	H	F	CH ₂ cPr	Ph-3-Et	63-64				
25	51	CF ₃	''	н	н	F	CH ₂ cPr	Ph-3-Me	52-53				
	52	CF ₃	H	н	н	F	CH ₂ cPr	Ph-3-Me-4-F	73-74				
	53	CF ₃	Н	н	H	F	CH ₂ cPr	Ph-3-Me-4-OMe	n ^{26.0} 1. 5307				
	54	CF ₃	H	Н	Н	F	CH ₂ cPr	Ph-4-F	58-59				
30	55	-	''	Н	Н	F	CH ₂ cPr	Ph-4-Me	n ^{26.0} 1.5248				
	56	CF ₃	^п н	Н	Н	F	_	Ph-4-OMe	75-76				
	57	CF ₃	H	Н	Н	F	CH ₂ cPr	Ph	72-74				
	58	CF ₃	Н	Н	Н.	F	CH ₂ cPr		72-74 54-56				
	59	CF ₃	Н				CH ₂ C≡CH	2-thienyl	56-58				
35		CF ₃	Н	H	Н	F	CH ₂ C≡CH	3-thienyl	57-58				
	60 61	CF ₃		H	Н	F	CH ₂ C≔CH	Ph-2-F	n ^{22.5} 1.5257				
	62	CF ₃	H	H	Н	F	CH ₂ C≡CH	Ph-2-F-4-OMe	n 1.5257 n ^{28:0} 1.5192				
	l 1	CF ₃	Н	Н :	Н	F	CH ₂ C≡CH	Ph-2-F-5-Me	98-100				
40	63	CF ₃	H	н	н	F	CH ₂ C≡CH	Ph-3,5-Me ₂	98-100 95-96				
,,,	64 65	CF ₃	Н	Н	Н	F	CH ₂ C=CH	Ph-4-Me	n ^{27.5} 1.5370				
	l i	CF ₃	H	H	Н	F	CH ₂ C≡CH	Ph-4-OMe	1 13 1				
	66	CF ₃	Н	H	Н	F	CH ₂ C≡CH	Ph	58-60 n ^{22.5} 1. 5557				
	67	CF₃	H	H	Н	F	CH ₂ C≡CI	Ph-4-OMe	n 1.5557 85-86				
45	68	CF ₃	н	Н	Н	F	iPr	Ph-4-OMe]				
	69	CF ₃	H	H	Н	F	iPr	Ph	84-85				
	70	CF ₃	Н	н	Н	F	nBu	Ph	n ^{18.5} 1.5121				
	71	CF ₃	Н	н	Н	F	nPr	Ph-2-F-4-OMe	n 1.5121				
50	72	CF ₃	Н	H	Н	F	nPr	Ph-2-F-5-Me	n ¹⁹⁰ 1. 5129				
50	73	CF ₃	Н	Н	H	F	nPr	Ph-4-Me	59-60				
	74	CF ₃	Н	н	Н	F -	nPr	Ph-4-OMe	54-55				
	75	CF ₃	Н	H	Н	F	nPr	Ph	n ^{26.5} 1. 5106				
	76	CF₃	Н	Н	Н	F	tBu	Ph					
55	77	CF ₃	Н	Н	Н	CI	CH ₂ CH=CH ₂	Ph-4-OMe					
	78	CF ₃	Н	Н	Н	CI	CH(CH ₃)CH=CH ₂	Ph-4-OMe					
	79	CF ₃	Н	Н	Н	CI	CH₂CH=CHCI	Ph-4-OMe					

Table 3. (continued)

	$(r^1, r^2 = H)$												
5	No.	X ¹	X ²	X ₃	X4	X5	R ¹	R²	Physical const. mp (°C) refractive index				
	80	CF ₃	Н	Н	Н	CI	CH ₂ CH=CCl ₂	Ph-4-OMe	-				
	81	CF ₃	н	Н	Н	CI	CH ₂ CH=CHCH ₃	Ph-4-OMe					
10		_		Н	1	CI	Et	İ	60.70				
,,,	82 83	CF ₃ CF ₃	H H	l H	H	CI	CH ₂ CH ₂ CI	Ph-4-OMe Ph-4-OMe	68-70				
	84	CF ₃	''	;;	''	CI	CH ₂ CHF ₂	Ph-4-OMe					
	85	CF ₃	''	;;	;;	CI	Me Me	Ph-4-OMe					
	86	CF ₃	;;	н	н	CI	CH ₂ C(CI)=CH ₂	Ph-4-OMe					
15	87	CF ₃	Н	H	Н	CI	CH ₂ C(CH ₃)=CH ₂	Ph-4-OMe					
	88	CF ₃	Н	H	н	CI	CH ₂ CN	Ph-4-OMe					
	89	CF ₃	н	н	н	CI	CH ₂ OMe	Pb-4-OMe					
	90	CF ₃	н	н	н	CI	CH ₂ cPr	3-methyl					
20							_	pyrazol-1-yl					
	91	CF ₃	н	Н	н	CI	CH ₂ cPr	4-methyl					
!						İ		pyrazol-1-yl					
	92	CF ₃	Н	Н	Н	CI	CH ₂ cPr	pyrazol-1-yl					
25	93	CF ₃	Н	Н	н	CI	CH ₂ cPr	3-methyl					
20	94	^F	ا ن	u	١.,		CU - D-	2-thienyl					
	94	CF ₃	Н	Н	Н	CI	CH₂cPr	4-methyl 2-thienyl					
	95	CF ₃	н	н	н	CI	CH ₂ cPr	5-methyl					
	55	O. 3	''		• • •	0.	0112011	2-thienyl					
30	96	CF ₃	н	н	н	CI	CH₂cPr	2-thienyl	47-49				
	97	CF ₃	Н	н	н	CI	CH ₂ cPr	4-methyl					
		•					_	3-thienyl					
	98	CF ₃	н	Н	Н	CI	CH₂cPr	5-methyl					
35								3-thienyl					
	99	CF ₃	Н	Н	Н	CI	CH ₂ cPr	3-thienyl	74-76				
	100	CF ₃	H	Н	H	CI	CH₂cPr	Ph-2,4-F ₂					
	101	CF ₃	H	Н	H	CI	CH ₂ cPr	Ph-2-F					
40	102	CF ₃	H	н	H	CI	CH ₂ cPr	Ph-2-F-3-Me					
	103 104	CF ₃	1 1	H	H	CI CI	CH ₂ cPr	Ph-2-F-4-OMe					
	104	CF₃		П	П		CH₂cPr	Ph-2-F-4-OMe 5-Me					
	105	CF ₃	н	н	н	CI	CH ₂ cPr	Ph-2-F-5-Me	n ^{23.8} 1.5323				
	106	CF ₃	H	H	н	CI	CH ₂ cPr	Ph-3,5-Me ₂	_D 1.0025				
45	107	CF ₃	н	Н	Н	CI	CH ₂ cPr	Ph-3-Me					
	108	CF ₃	н	Н	н	CI	CH ₂ cPr	Ph-3-Me-4-F					
	109	CF ₃	н	н	н	CI	CH ₂ cPr	Ph-3-Me-4-OMe					
	110	CF ₃	н	н	Н	CI	CH ₂ cPr	Ph-4-F					
50	111	CF ₃	н	н	н	CI	CH₂cPr	Ph-4-Me					
	112	CF ₃	н	н	н	CI	CH ₂ cPr	Ph-4-OMe	65-67				
	113	CF ₃	н	н	Н	CI	CH₂cPr	Ph	63-65				
	114	CF ₃	н	н	н	CI	CH ₂ C≡CH	Ph-4-OMe	76-78				
55	115	CF ₃	н	н	н	CI	CH ₂ C=CH	Ph	96-97				
	116	CF ₃	Н	Н	Н	CI	CH ₂ C=CI	Ph-4-OMe					
	117	CF ₃	Н	н	Н	CI	iPr	Ph-4-OMe					

Table 3. (continued)

	(r¹, r² =H)												
	No.	X ¹	X ²	X3	X4	X5	, 	R ²	Dhariadaaa				
5	INO.	^'	^-	^		**	R ¹	H²	Physical const. mp (°C) refractive index				
	118	CF ₃	H	Н	Н	CI	nBu	Ph-4-OMe					
	119	CF ₃	н	Н	н	CI	nPr	Ph-4-OMe					
10	120	CF ₃	н	Н	н	CI	tBu	Ph-4-OMe					
	121	CF ₃	н	''	H	CF ₃	CH ₂ CH=CH ₂	Ph-4-OMe					
	122	CF ₃	Н	Н	H	CF ₃	CH(CH ₃)CH=CH ₂	Ph-4-OMe					
	123	CF ₃	н	Н	н	CF ₃	CH ₂ CH=CHCI	Ph-4-OMe					
	124	CF ₃	Н	Н	Н	CF ₃	CH ₂ CH=CCl ₂	Ph-4-OMe					
15	125	CF ₃	Н	н	н	CF ₃	CH₂CH=CHCH ₃	Ph-4-OMe					
	126	CF ₃	н	н	н	CF ₃	Et	Ph-4-OMe					
	127	CF ₃	Н	Н	н	CF ₃	CH ₂ CH ₂ CI	Ph-4-OMe					
	128	CF ₃	н	н і	н	CF ₃	CH ₂ CHF ₂	Ph-4-OMe					
20	129	CF ₃	н	н	н	CF ₃	Me	Ph-4-OMe					
	130	CF ₃	Н	н	Н	CF ₃	CH ₂ C(CI)=CH ₂	Ph-4-OMe					
	131	CF ₃	н	н	н	CF ₃	CH ₂ C(CH ₃)=CH ₂	Ph-4-OMe					
	132	CF ₃	H ,	н	Н	CF ₃	CH ₂ CN	Ph-4-OMe					
05	133	CF ₃	Н	н	н	CF ₃	CH ₂ OMe	Ph-4-OMe					
25	134	CF3	Н	н	Н	CF ₃	CH ₂ cPr	3-methyl					
								pyrazol-1-yl					
	135	CF ₃	н	Н	Н	CF ₃	CH ₂ cPr	4-methyl-					
	136	CF ₃	Н	Н	Н	CF ₃	CH₂cPr	-pyrazol-1-yi Ph					
30	137	CF ₃	Н	Н	Н	CF ₃	CH₂cPr	pyrazol-1-yl					
								3-methyl					
	138	C E	ا با	н	1.1	~-	CI -D-	2-thienyl					
	136	CF ₃	Н		н	CF ₃	CH ₂ cPr	4-methyl 2-thienyl					
35	139	CF ₃	н	н	н	CF ₃	CH ₂ cPr	5-methyl					
	100	0.3		· '		0,3	3/120/1	2-thienyl					
	140	CF ₃	н	н	н	CF ₃	CH ₂ cPr	2-thienyl	n ^{22.4} 1.5082				
	141	CF ₃	н	н	н	CF ₃	CH ₂ cPr	4-methyl	D				
		•					-	3-thienyl					
40	142	CF ₃	н	н	н	CF ₃	CH ₂ cPr	5-methyl					
								3-thienyl					
	143	CF ₃	н	н	Н	CF ₃	CH ₂ cPr	3-thienyl					
	144	CF ₃	н	н	н	CF ₃	CH ₂ cPr	Ph-2,4-F ₂					
45	145	CF ₃	н	н	Н	CF ₃	CH ₂ cPr	Ph-2-F					
	146	CF ₃	н	Н	Н	CF ₃	CH ₂ cPr	Ph-2-F-3-Me					
	147	CF ₃	н	н	Н	CF ₃	CH ₂ cPr	Ph-2-F-4-OMe					
	148	CF ₃	н	н	Н	CF ₃	CH₂cPr	Ph-2-P-4-OMe					
50	149	CE	L.	н	u	CE	CM aD=	5-Me	74.75				
-	150	CF ₃ CF ₃	H	H	H	CF ₃ CF ₃	CH ₂ cPr	Ph-2-F-5-Me	74 -75				
	151	CF ₃	H	Н	Н	CF ₃	CH ₂ cPr	Ph-3,5-Me ₂ Ph-3-Me					
	152	CF ₃	Н	н	Н	CF ₃	CH ₂ cPr CH ₂ cPr	Pn-3-Me Ph-3-Me-4-F					
	153	CF ₃	Н	н	Н	CF ₃	CH ₂ cPr	Ph-3-Me-4-OMe					
55	154	CF ₃	н	н	н	CF ₃	CH ₂ cPr	Ph-4-F					
	155	- 1	н	н	н	_	=	Ph-4-Me					
l		CF ₃				CF ₃	CH ₂ cPr	111 7 1110					

Table 3. (continued)

	(r¹, r² =H)												
	No.	X1	X2	X ₃	X4	X5	R1	R2	I Dhuaisat as a Ausa				
5	100.	*	^-	**	**		H'	H ²	Physical const. mp (°C) refractive index				
	156	CF ₃	Н	H	Н	CF ₃	CH ₂ cPr	Ph-4-OMe	n ^{22.6} 1.5031				
	157	CF ₃	н	Н	н	CF ₃	CH ₂ cPr	Ph	68-69				
10	158	CF ₃	Н	Н	н	CF ₃	CH ₂ C≡CH	Ph-4-OMe					
	159	CF ₃	Н	н	Н	CF ₃	CH ₂ C≡CI	Ph-4-OMe					
	160	CFa	Н	Н	H	CF ₃	iPr	Ph-4-OMe					
	161	CF ₃	Н	H	H	CF ₃	nBu	Ph-4-OMe					
45	162	CF ₃	н	Н	Н	CF ₃	nPr	Ph-4-OMe					
15	163	CF ₃	Н	Н	Н	CF ₃	tBu	Ph-4-OMe					
	164	CF ₃	н	Н	н	OMe	CH ₂ CH=CH ₂	Ph-4-OMe					
	165	CF ₃	Н	Н	н	OMe	CH(CH ₃)CH=CH ₂	Ph-4-OMe					
	166	CF ₃	H	Н	н	OMe	CH2CH=CHCI	Ph-4-OMe					
20	167	CF ₃	н	Н	H	OMe	CH ₂ CH=CCl ₂	Ph-4-OMe					
	168	CF ₃	Н	н	н	OMe	CH2CH=CHCH3	Ph-4-OMe					
	169	CF ₃	Н	Н	Н	OMe	Et	Ph-4-OMe	95-96				
	170	CF ₃	н	Н	Н	OMe	CH ₂ CH ₂ CI	Ph-4-OMe					
25	171	CF ₃	Н	Н	н	OMe	CH ₂ CHF ₂	Ph-4-OMe					
23	172	CF ₃	H	Н	н	OMe	Me	Ph-4-OMe					
	173	CF ₃	H	Н	Н	OMe	CH ₂ C(CI)=CH ₂	Ph-4-OMe					
	174	CF ₃	Н	Н	Н	OMe	CH ₂ C(CH ₃)=CH ₂	Ph-4-OMe	-				
	175	CF ₃	н	Н	H	OMe	CH ₂ CN	Ph-4-OMe					
30	176	CF ₃	Н	H	Н	OMe	CH ₂ OMe	Ph-4-OMe					
	177	CF ₃	Н	H	Н	OMe	CH ₂ cPr	3-methyl					
				١				pyrazol-1-yi					
	178	CF ₃	Н	Н	Н	OMe	CH ₂ cPr	4-methyl					
35	179	CE	ا ا	Н	H	014-	OU - B-	pyrazol-1-yl					
	180	CF ₃ CF ₃	H H		Н	OMe	CH ₂ cPr	pyrazol-1-yl					
	180	CF3	"	"	П	OMe	CH ₂ cPr	3-methyl 2-thienyl	1				
	181	CF ₃	Н	н	н	OMe	CH ₂ cPr	4-methyl]				
		3	, ,		''		0112011	2-thienyl					
40	182	CFa	н	н	Н	OMe	CH₂cPr	5-methyl					
		•					22	2-thienyl					
	183	CF ₃	н	н	н	OMe	CH₂cPr	2-thienyl					
							_	4-methyl-					
45	184	CF ₃	н	н	Н	OMe	CH₂cPr	-3-thienyl					
	185	CF ₃	н	Н	Н	OMe	CH₂cPr	5-methyl					
								3-thienyl					
	186	CF ₃	Н	Н	Н	OMe	CH ₂ cPr	3-thienyl					
50	187	CF ₃	Н	Н	Н	OMe	CH ₂ cPr	Ph-2,4-F ₂					
50	188	CF ₃	H	Н	H	OMe	CH₂cPr	Ph-2-F					
	189	CF ₃	Н	H	H	OMe	CH₂cPr	Ph-2-F-3-Me					
	190	CF ₃	н	H	Н	OMe	CH₂cPr	Ph-2-F-4-OMe					
	191	CF ₃	Н	Н	Н	OMe	CH ₂ cPr	Ph-2-F-4-OMe					
55	192	CE.	_{[1}	ایا	Li	014-	011.5	5-Me					
	193	CF ₃	H	Н	H	OMe	CH₂cPr	Ph-2-F-5-Me	}				
	133	CF ₃	17	Н	FI	OMe	CH ₂ cPr	Ph-3,5-Me ₂					

Table 3. (continued)

	$(r^1, r^2 = H)$												
5	No.	X ¹	X²	X3	X ⁴	X5	R ¹	R ²	Physical const. mp (°C) refractive index				
	194	CF ₃	Н	Н	Н	OMe	CH ₂ cPr	Ph-3-Me					
		-	н	Н	н	OMe	_	Ph-3-Me-4-F					
	195	CF ₃			1		CH ₂ cPr	F11-3-146-4-F					
10	196	CF3	н	Н	Н	OMe	CH₂cPr	Ph-3-Me-4-OMe					
	197	CF ₃	н	Н	Н	OMe	CH₂cPr	Ph-4-F					
	198	CF ₃	н	Н	Н	OMe	CH ₂ cPr	Ph-4-Me					
	199	CF3	Н	Н	Н	OMe	CH ₂ cPr	Ph-4-OMe	92-93				
15	200	CF ₃	н	Н	H	OMe	CH ₂ cPr	Ph-3-Cl-4-OMe					
	201	CF ₃	H	Н	н	OMe	CH₂cPr	Ph-3-Cl-4-OMe 5-Me					
	202	CF ₃	Н	Н	H	OMe	CH ₂ cPr	Ph					
	203	CF ₃	Н	Н	Н	OMe	CH ₂ C≡CH	Ph-4-OMe					
20	204	CF ₃	Н	Н	н	OMe	CH ₂ C≡CI	Ph-4-OMe					
	205	CF ₃	Н	Н	н	OMe	iPr	Ph-4-OMe					
	206	CF ₃	H	Н	Н	OMe	nBu	Ph-4-OMe					
	207	CF ₃	Н	н	Н	OMe	nPr	Ph-4-OMe					
	208	CF ₃	н	н	Н	OMe	tBu	Ph-4-OMe					
25	209	CF ₃	Н	Н	Н	SMe	Et	Ph-4-OMe	89-90				
	210	CF ₃	н	Н	Н	SMe	Me	Ph-4-OMe					
	211	CF ₃	Н	Н	Н	SMe	CH₂cPr	Ph-4-OMe	73-75				
	212	CF ₃	Н	Н	Н	SMe	CH ₂ cPr	Ph	106-109				
30	213	CF ₃	н	Н	н	SMe	nBu	Ph-4-OMe					
	214	CF ₃	н	Н	н	SMe	tBu	Ph-4-OMe					
	215	CF ₃	н	н	CI	н	CH ₂ CH=CH ₂	Ph-4-OMe					
	216	CF ₃	н	н	CI	Н	Et	Ph-4-OMe					
	217	CF3	н	н	CI	Н	Me	Ph-4-OMe					
35	218	CF ₃	н	н	CI	н	CH₂cPr	3-methyl pyrazol-1-yl					
	219	CF ₃	н	н	CI	Н	CH ₂ cPr	4-methyl					
	ŀ							pyrazol-1-yl					
40	220	CF ₃	н	н	CI	Н	CH ₂ cPr	pyrazol-1-yl					
	221	CF ₃	Н	Н	CI	Н	CH₂cPr	3-methyl					
								2-thienyl					
	222	CF ₃	Н	Н	CI	Н	CH ₂ cPr	4-methyl					
								2-thlenyl					
45	223	CF ₃	н	Н	CI	н	CH₂cPr	5-methyl					
								2-thienyl					
	224	CF ₃	Н	Н	CI	Н	CH ₂ cPr	2-thienyl					
	225	CF ₃	Н	Н	CI	н	CH ₂ cPr	4-methyl					
50		0.5		l l	01		OLL - D.	3-thienyl					
50	226	CF ₃	Н	Н	CI	Н	CH ₂ cPr	5-methyl					
	227	CF ₃	н	н	CI	н	CH ₂ cPr	3-thienyl					
	228	CF ₃	Н	Н	CI	Н	CH ₂ cPr	3-thienyl Ph-4-OMe					
	228		Н	Н	CI	Н	CH ₂ CPr CH ₂ C ≡CH	Ph-4-OMe Ph-4-OMe					
55	230	CF ₃	Н	Н	CI	Н	i CH ₂ C ≡CH iPr	Pn-4-OMe Ph-4-OMe					
	230	CF ₃	Н	C H	CI	Н	nBu	Ph-4-OMe Ph-4-OMe					
	231	CF ₃	11	- 17	<u> </u>		IIDU	FII-4-OIVIE					

Table 3. (continued)

	(r¹, r² =H)											
5	No.	X ¹	X2	Χ3	X ⁴	X5	R ¹	R ²	Physical const. mp (°C) refractive index			
	232	CF ₃	Н	H	CI	Н	nPr	Ph-4-OMe				
	233	_	н	н	CI	Н	tBu	Ph-4-OMe				
10		CF ₃					ł					
10	234	CF ₃	H	Н	CI	F	CH ₂ CH=CH ₂	Ph-4-OMe	81-83			
	235 236	CF ₃	H	H	CI	F	CH(CH ₃)CH=CH ₂	Ph-4-OMe				
	237	CF ₃ CF ₃	H H	H H	CI CI	F	CH ₂ CH=CHCI	Ph-4-OMe				
	238	CF ₃		Н	CI	F	CH ₂ CH=CCl ₂	Ph-4-OMe Ph-4-OMe				
15	239	CF ₃	''	H	CI	F	CH ₂ CH=CHCH ₃ Et	Ph-4-OMe	99- 100			
	240	CF ₃	н	Н	Ci	F	CH ₂ CH ₂ CI	Ph-4-OMe	n ^{24.0} 1.5301			
	241	CF ₃	:: н	н	Ci	F	CH ₂ CH ₂ CI	Ph	90-91			
	242	CF ₃	H	;;	Ci	F	CH ₂ CH ₂ CI	Ph-2-F-5-Me	n ^{24.0} 1.5241			
20	243	CF ₃	н	Н	CI	F	CH ₂ CHF ₂	Ph-4-OMe				
	244	CF ₃	н	н	CI	F	Me	Ph-4-OMe				
	245	CF ₃	н	Н	CI	F	CH ₂ C(CI)=CH ₂	Ph-4-OMe				
	246	CF ₃	Н	н	CI	F	CH ₂ C(CH ₃)=CH ₂	Ph-4-OMe				
25	247	CF ₃	Н	н	CI	F	CH ₂ CN	Ph-4-OMe				
25	248	CF ₃	н	н	CI	F	CH ₂ OMe	Ph-4-OMe				
	249	CF ₃	Н	Н	CI	F	CH₂cPr	3-methyl				
						ļ		pyrazol-1-yl				
	250	CF ₃	H	Н	CI	F	CH ₂ cPr	4-methyl				
30	054	05			01	_	OLL D	pyrazol-1-yl				
	251 252	CF ₃	H	Н	CI	F	CH ₂ cPr	pyrazol-1-yl				
	252	CF ₃	"	Н	CI	F	CH₂cPr	3-methyl 2-thienyl				
	253	CF ₃	Н	н	CI	F	CH₂cPr	4-methyl				
35		U . 9	,,,		0.	•	0112011	2-thienyl				
	254	CF ₃	н	н	CI	F	CH₂cPr	5-methyl				
		J	1				-	2-thienyl				
	255	CF ₃	н	н	CI	F	CH₂cPr	2-thienyl	112-113			
40	256	CF ₃	н	н	CI	F	CH ₂ cPr	4-methyi				
40								3-thienyl				
	257	CF ₃	н	Н	CI	F	CH ₂ cPr	5-methyl				
	252	<u> </u>	ا ا		~ ;	_	011.5	3-thienyl				
	258	CF₃	Н	н	CI	F	CH₂cPr	3-thienyl	125-126			
45	259	CF ₃	H	Н	CI	F	CH ₂ cPr	Ph-2,4-F ₂	57.50			
	260 261	CF ₃	Н	Н	CI	F	CH₂cPr	Ph-2-F	57-59			
	262	CF ₃	H	н	CI	F	CH ₂ cPr	Ph-2-F-3-Me	n ^{26.7} 1.5200			
	263	CF ₃	Н	Н	CI	F	CH ₂ cPr	Ph-2F-4-OMe Ph-2-F-4-OMe	73-74			
50	200	Org		п]	OI	F	CH₂cPr	Pn-2-F-4-ОМө 5-Ме	/3-/4			
	264	CF3	н	н	CI	F	CH ₂ cPr	Ph-2-F-5-Me	60-62			
	265	CF ₃	н	н	CI	F	CH ₂ cPr	Ph-3,5-Me ₂	55 52			
	266	CF ₃	н	н	CI	F	CH ₂ cPr	Ph-3-Me	92-93			
	267	CF ₃	н	н	CI	F	CH ₂ cPr	Ph-3-Me-4-F	98-99			
55	268	CF ₃	н	Н	CI	F	CH ₂ cPr	Ph-3-Me-4-OMe	86-87			
	269	CF ₃	н	н	CI	F	. CH ₂ cPr	Ph-4-F	94-95			
		3					. 51 1201 1					

Table 3. (continued)

	$(r^1, r^2 = H)$												
5	No.	X ¹	X2	Хз	X4	X ⁵	R ¹	R²	Physical const. mp (°C) refractive index				
	270	CF ₃	H	Н	CI	F	CH ₂ cPr	Ph-4-Me	75-76				
	271	CF ₃	н	н	CI	F	CH ₂ cPr	Ph-4OMe	98				
10	272	CF ₃	н	Н	CI	F	CH ₂ cPr	Ph	112				
	273	CF ₃	Н	н	Ci	F	CH ₂ C≡CH	Ph	54-58				
	274	CF ₃	н	Н	CI	F	CH ₂ C≡CH	Ph-2-F-5-Me	91-92				
	275	CF ₃	H	н	CI	F	CH ₂ C≔CH	Ph-4-OMe	102-103				
15	276	CF ₃	Н	н	CI	F	CH ₂ C≕CI	Ph-4-OMe	112-114				
15	277	CF ₃	Н	н	CI	F	iPr	Ph-4-OMe	124-125				
	278	CF3] н	н	CI	F	nBu	Ph-4-OMe]				
	279	CF ₃	Н	Н	CI	F	nPr	Ph-4-OMe	90-91				
	280	CF ₃	Н	н	CI	F	tBu	Ph-4-OMe					
20	281	CF ₃	н	н	CI	CI	CH₂CH=CH₂	Ph-4-OMe	80-81				
	282	CF ₃	н	н	CI	CI	CH(CH ₃)CH=CH ₂	Ph-4-OMe					
	283	CF ₃	Н	Н	CI	CI	CH2CH=CHCI	Ph-4-OMe					
	284	CF ₃	н	н	CI	CI	CH₂CH=CCl₂	Ph-4-OMe					
	285	CF ₃	Н	н	CI	CI	CH₂CH=CHCH₃	Ph-4-OMe	. :				
25	286	CF ₃	н	н	CI	CI	Et	Ph	123-124				
	287	CF ₃	н	н	CI	CI	Et	Ph-2-F-5-Me	78-81				
	288	CF ₃	н	Н	CI	CI	Et	Ph-4-OMe	90-91				
	289	CF ₃	н	н	CI	CI	CH ₂ CH ₂ CI	Ph	104-105				
30	290	CF ₃	н	н	CI	CI	CH ₂ CH ₂ CI	Ph-2-F-5-Me	n ^{26.5} 1.5309				
	291	CF3	н	н	CI	CI	CH ₂ CH ₂ CI	Ph-4-OMe					
	292	CF3	н	н	CI	CI	CH ₂ CHF ₂	Ph-4-OMe					
	293	CF3	н	н	CI	CI	Me	Ph-4-OMe	77-79				
	294	CF ₃	н	н	CI	CI	CH ₂ C(CI)=CH ₂	Ph-4-OMe					
35	295	CF ₃	Н	н	CI	CI	CH ₂ C(CH ₃)=CH ₂	Ph-4-OMe					
	296	CF ₃	Н	н	CI	CI	CH ₂ CN	Ph-4-OMe					
	297	CF ₃	н	Н	CI	CI	CH ₂ OMe	Ph-4-OMe					
	298	CF ₃	н	Н	CI	CI	CH₂cPr	3-methyl					
40							_	pyrazol-1-yl					
	299	CF ₃	н	н	CI	CI	CH ₂ cPr	4-methyl					
								pyrazol-1-yl					
	300	CF ₃	Н	Н	CI	CI	CH ₂ cPr	pyrazol-1-yl					
	301	CF ₃	н	н	CI	CI	CH ₂ cPr	3-methyl					
45								2-thienyl					
	302	CF ₃	н	Н	CI	CI	CH ₂ cPr	4-methyl					
								2-thienyl					
	303	CF ₃	Н	н	CI	CI	CH₂cPr	5-methyl					
50	204	<u> </u>	١		~ !	01	OU - D	2-thienyl	400 444				
	304 305	CF ₃	Н	Н	CI CI	CI	CH₂cPr	2-thienyl	139-141				
	305	CF ₃	н	H	OI	CI	CH₂cPr	4-methyl 3-thienyl					
	306	CF ₃	н	н	CI	CI	CH 20*	-					
	555	U:3	''	-''-	OI	OI	CH₂cPr	5-methyl 3-thienyl					
55	307	CF ₃	н	н	CI	CI	CH ₂ cPr	3-thienyl	140-141				
	308	- 1	н	н]	CI	CI	- 1	-	170 171				
į		CF ₃					CH ₂ cPr	Ph-2,4-F ₂					

Table 3. (continued)

	$(r^1, r^2 = H)$											
5	No.	X ¹	X2	Χ³	X ⁴	X5	R ¹	R ²	Physical const. mp (°C) refractive index			
	309	CF ₃	н	н	CI	CI	CH ₂ cPr	Ph-2-F	n ^{23.0} 1.5404			
	310	_	Н	Н	CI	CI	_	Ph-2-F-3-Me	" p 1.5464			
10	311	CF ₃			İ		CH ₂ cPr		n ^{22.5} 1.5371			
70		CF₃	H	H	CI	CI	CH ₂ cPr	Ph-2-F-4-OMe	1 AD#			
	312	CF ₃		Н	CI	CI	CH₂cPr	Ph-2F-4-OMe 5-Me	n ^{29.5} 1.5287			
	313	CF ₃	Н	Н	CI	CI	CH ₂ cPr	Ph-2-F-5-Me	56-57			
	314	CF ₃	н	;;	CI	CI	CH ₂ cPr	Ph-3,5-Me ₂	30-57			
15	315	CF ₃	Н	::	CI	CI	CH ₂ cPr	Ph-3-Me	111-114			
	316	CF ₃	lн	н	CI	CI	CH ₂ cPr	Ph-3-Me-4-F	113-114			
	317	CF ₃	Н	Н	CI	CI	CH ₂ cPr	Ph-3-Me-4-OMe	710 114			
	318	CF ₃	Н	H	Ci	CI	CH ₂ cPr	Ph-4-F				
20	319	CF ₃	Н	н	Ci	CI	CH ₂ cPr	Ph-4-Me	93-95			
	320	CF ₃	н	Н	CI	CI	CH ₂ cPr	Ph-4-OMe	97-98			
	321	CF ₃	Н	Н	CI	CI	CH ₂ cPr	Ph	134-135			
	322	CF ₃	н	н	CI	CI	CH₂C≔CH	Ph	96-98			
	323	CF ₃	н	н	CI	Cı	CH ₂ C≡CH	Ph-2-F-5-Me	77-79			
25	324	CF ₃	н	н	CI	CI	CH ₂ C≡CH	Ph-4-OMe	94-96			
	325	CF ₃	Н	н	CI	CI	CH ₂ C=CI	Ph-4-OMe				
	326	CF ₃	Н	н	CI	CI	iPr	Ph-4-OMe	123-124			
	327	CF ₃	н	н	CI	CI	nBu	Ph-4-OMe				
30	328	CF ₃	н	н	CI	CI	nPr	Ph-4-OMe	88-89			
	329	CF ₃	H	Н	CI	CI	tBu	Ph-4-OMe				
	330	CF ₃	н	н	SMe	F	CH ₂ cPr	Ph-4-OMe	114-115			
	331	CF ₃	Н	н	OMe	F	CH ₂ cPr	Ph-4-OMe	125-126			
	332	CF ₃	н	н	OMe	F	CH ₂ cPr	Ph-4-OMe	135-136			
35	333	CF ₃	Н	н	Me	F	CH ₂ cPr	Ph-4-OMe	71-72			
	334	CF ₃	Н	F	Н	н	Et	Ph-4-OMe	70-72			
- 1	335	CF ₃	н	F	Н	н	Me	Ph-4-OMe				
	336	CF ₃	н	F	Н	н	CH₂cPr	Ph-4-OMe	97-99			
40	337	CF ₃	Н	F	Н	Н	CH₂cPr	Ph	117-119			
	338	CF ₃	Н	F	Н	Н	nBu	Ph-4-OMe				
	339	CF ₃	Н	F	Н	H	tBu	Ph				
	340	CF ₃	Н	CI	Н	Н	Et	Ph-4-OMe	89-91			
45	341	CF ₃	Н	CI	Н	Н	Me	Ph-4-OMe				
45	342	CF ₃	Н	CI	Н	Н	CH ₂ cPr	Ph-4-OMe	97-99			
	343	CF ₃	Н	CI	Н	н	CH ₂ cPr	Ph	91-92			
	344	CF ₃	Н	CI	Н	Н	nBu	Ph-4-OMe				
l	345	CF ₃	Н	CI	Н	H	tBu	Ph				
50	346	CF ₃	CI	н	Н	F	CH ₂ cPr	Ph-4-OMe	Oii, ‡l			
	347	CF ₃	CI	Н	Н	CI	CH ₂ cPr	2-thienyl	74			
l	348	CF ₃	CI	н	Н	CI	CH ₂ cPr	Ph-4-OMe	100			
İ	349	CF ₃	CI	н	Н	CI	CH ₂ cPr	Ph	95-96			
_	350	CF ₃	H	н	F	F	CH ₂ CH=CH ₂	Ph-4-OMe				
55	351	CF ₃	H	Н	F	F	Et	Ph	81-83			
	352	CF ₃	H	H	F	F	Et	Ph-2-F-5-Me	79-80			
	353	CF ₃	Н	Н	F	F	Et	Ph-4-OMe	78-79			

Table 3. (continued)

	(r¹, r² =H)											
_	No.	X ¹	X2	X ₃	X4	X5	R ¹	H2	Physical const. mp			
5									(°C) refractive index			
	354	CF ₃	Н	Н	F	F	CH ₂ CH ₂ CI	Ph	n ^{23.5} 1.5245			
	355	CF ₃	Н	Н	F	F	CH ₂ CH ₂ CI	Ph-2-F-5-Me	D 1.5240			
10	356	CF ₃	н	Н	F	F	CH ₂ CH ₂ CI	Ph-4-OMe				
,,,	357	CF ₃	Н	Н	F	F	Me	Ph-4-OMe				
	358	CF ₃	Н	Н	F	F	CH ₂ OMe	Ph-4-OMe				
	359	CF ₃	н	Н	F	P	CH ₂ cPr	3-methyl				
		- 3					•	pyrazol-1-yl				
15	360	CF ₃	Н	Н	F	F	CH ₂ cPr	4-methyl				
				l				pyrazol-1-ył				
	361	CF ₃	Н	Н	F	F	CH₂cPr	pyrazol-1-yi				
	362	CF ₃	Н	н	F	F	CH ₂ cPr	3-methyl				
		•	ŀ				_	2-thienyl				
20	363	CF ₃	Н	Н	F	F	CH₂cPr	4-methyl				
								2-thienyl	ļ			
	364	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-2-F	56-58			
	365	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-2-F-3-Me				
25	366	CF ₃	Н	Н	F	F	CH₂cPr	Ph-2F-4-OMe				
	367	CF ₃	Н	Н	F	F	CH₂cPr	Ph-2F-4-OMe 5-Me	70-73			
	368	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-2F-5-Me	67-69			
	369	CF ₃	Н	н	F	F	CH ₂ cPr	Ph-3,5-Me ₂				
30	370	CF ₃	н	н	F	F	CH ₂ cPr	Ph-3-Me	70-72			
	371	CF3	н	Н	F	F	CH ₂ cPr	Ph-3-Me-4-F	51-53			
	372	CF ₃	н	Н	F	F	CH ₂ cPr	Ph-3-Me-4-OMe	56-57			
	373	CF3	н	н	F	F	CH ₂ cPr	Ph-4-F	70-72			
35	374	CF ₃	н	н	F	F	CH ₂ cPr	Ph-4-Me	64-66			
35	375	CF ₃	н	н	F	F	CH ₂ cPr	Ph-4-OMe	73-74			
	376	CF ₃	н	н	F	F	CH ₂ cPr	Ph	61-62			
	377	CF ₃	н	н	F	F	CH ₂ C≡CH	Ph	76-78			
	378	CF ₃	н	н	F	F	CH ₂ C≡CH	Ph-2-F-5-Me	84-86			
40	379	CF ₃	н	н	F	F	CH ₂ C≡CH	Ph-4-OMe	100-102			
	380	CF ₃	н	н	F	F	CH ₂ C≡CI	Ph-4-OMe				
	381	CF ₃	н	н	F	F	iPr	Ph-4-OMe	65-66			
	382	CF ₃	Н	н	F	F	nBu	Ph-4-OMe				
	383	CF ₃	н	н	F	F	nPr	Ph-4-OMe	67-69			
45	384	CF ₃	н	н	F	F	tBu	Ph-4-OMe				
	385	CF ₃	н	н	F	F	CH₂cPr	5-methyl 2-thienyl				
	386	CF ₃	н	н	· F	F	CH ₂ cPr	2-thienyl	69-71			
50	387	CF ₃	Н	н	F	F	CH ₂ cPr	4-methyl				
30							_	3-thienyl				
	388	CF ₃	Н	н	F	F	CH ₂ cPr	5-methyl 3-thienyl				
	389	CF ₃	Н	Н	F	F	CH ₂ cPr	3-thienyl	79-81			
55	390	CF ₃	н	н	F	F	CH ₂ cPr	Ph-2,4-F ₂				
	391	CF ₃	н	Н	CI	F	nPr	Ph	113-114			
	392	CF ₃	Н	Н	CI	F	· nPr	Ph-2-F-5-Me	60-61			

Table 3. (continued)

	$(r^1, r^2 = H)$										
5	No.	X ¹	X2	Хз	X4	Х5	R ¹	R ²	Physical const. mp (°C) refractive index		
	393	CF ₃	H	Н	CI	F	iPr	Ph	111-112		
	394	CF ₃	Н	Н	CI	F	iPr	Ph-2-F-5-Me			
40	395	CF ₃	;;	Н	CI	F	CH ₂ CH=CH ₂	Ph	107-108		
10	396	CF ₃	н	Н н	CI	F	CH ₂ CH=CH ₂	Ph-2-F-5-Me	68-70		
	397	CF ₃	Н	н	CI	F	CH ₃	Ph	00-70		
	398	CF ₃	н	н	CI	F	CH ₃	Ph-2-F-5-Me			
	399	CF ₃	H	;;	CI	F	CH ₂ C=CI	Ph	90-91		
15	400	CF ₃	Н	H	CI	F.	CH ₂ C≡Cl	Ph-2-F-5-Me	n ^{25.0} 1.5491		
	401	CF ₃	Н	Н	CI	F	CH ₂ CH ₃	Ph	101-102		
	402	CF ₃	Н	Н	CI	F	CH ₂ CH ₃	Ph-2-F-5-Me	60-62		
	403	CF ₃	н	lн	CI	CI	nPr	Ph	128-129		
20	404	CF ₃	н	Н	Ci	CI	nPr	Ph-2-F-5-Me	n ^{24.0} 1.5212		
40	405	CF ₃	н	Н	Ci	CI	iPr	Ph	125-127		
	406	CF ₃	н	Н	CI	CI	iPr	Ph-2-P-5-Me	n ^{24.7} 1.5245		
	407	CF ₃	Н	н	CI	CI	CH ₂ CH=CH ₂	Ph	115-116		
	408	CF3	н	Н	CI	CI	CH2CH=CH2	Ph-2-F-5-Me	62-63		
25	409	CF ₃	Н	Н	CI	CI	CH ₃	Ph	111-113		
	410	CF ₃	Н	Н	CI	CI	CH ₃	Ph-2-P-5-Me	80-81		
	411	CF ₃	н	Н	CI	CI	CH ₂ C≡CI	Ph			
	412	CF ₃	Н	Н	CI	CI	CH ₂ C≕CI	Ph-2-F-5-Me			
30	413	CF ₃	Н	Н	F	F	nPr	Ph	56-58		
	414	CF ₃	н	Н	F	F	nPr	Ph-2-F-5-Me	38-40		
	415	CF ₃	Н	н	F	F	iPr	Ph	81-82.5		
	416	CF ₃	Н	Н	F	F	iPr	Ph-2-F-5-Me			
	417	CF ₃	Н	Н	F	F	CH ₂ CH=CH ₂	Ph			
35	418	CF ₃	Н	Н	F	F	CH ₂ CH=CH ₂	Ph-2-F-5-Me			
	419	CF ₃	Н	н	F	F	CH₃	Ph			
	420	CF ₃	Н	Н	F	F	CH ₃	Ph-2-F-5-Me			
	421	CF ₃	H	Н	F	F	CH ₂ C≡Cl	Ph			
40	422	CF ₃	Н	Н	F	F	CH ₂ C≡CI	Ph-2-F-5-Me			
	423	CF ₃	Н	Н	F	CI	CH ₂ CH=CH ₂	Ph			
	424	CF ₃	Н	Н	F `	CI	CH ₂ CH=CH ₂	Ph-4-OMe			
	425	CF ₃	H	Н	F	CI	CH ₂ CH=CH ₂	Ph-2-F-5-Me			
45	426	CF ₃	Н	H	F	CI	CH₂CH₃	Ph Ph			
45	427 428	CF ₃	H	Н	F	CI	CH₃CH₃	Ph-4-OMe			
	429	CF ₃	H	Н	F	CI	CH ₂ CH ₃	Ph-2-F-5-Me	n ^{24.5} 1.5344		
	430	CF ₃	Н	H	F	CI	CH ₂ CH ₂ CI	Ph	n 1.5344		
	430	CF ₃	H	Н	F	CI	CH ₂ CH ₂ CI	Ph-4-OMe	n ^{29.5} 1.5294		
50	432	CF ₃ CF ₃	Н	H	F	Ci Ci	CH ₂ CH ₂ CI	Ph-2-F-5-Me			
	433	CF ₃	Н	H	F	CI	CH₃	Ph Ph-4-OMe			
	434	CF ₃	Н	Н	F	CI	CH₃ CH₃				
	435	CF ₃	H	Н	F	CI		Ph-2-F-5-Me Ph-4-OMe			
55	436	_	Н	Н	F	CI	CH ₂ OMe	3-methyl			
<i>33</i>	730	CF ₃	,,,	17	r-	O	CH ₂ cPr	pyrazol-1-yl			
			L					Pyrazor r-yr			

Table 3. (continued)

							(r ¹ , r ² =H)		
5	No.	X ¹	X²	Χ3	X4	X ⁵	R ¹	R²	Physical const. mp (°C) refractive index
	437	CF ₃	Н	Н	F	CI	CH ₂ cPr	4-methyl pyrazol-1-yl	
	438	CF3	н	н	· F	CI	CH ₂ cPr	pyrazol-1-yl	
10	439	CF ₃	н	н	F	CI	CH ₂ cPr	3-methyl	
			İ					2-thienyl] ,
	440	CF ₃	Н	Н	F	CI	CH ₂ cPr	4-methyl	
				١				2-thienyl	
15	441	CF₃	Н	Н	F	CI	CH ₂ cPr	5-methyl	
	442	CF ₃	Н	Н	F	CI	CH ₂ cPr	2-thienyl 2-thienyl	104-106
	443	CF ₃	''	Н	F	Ci	CH ₂ cPr	4-methyl	104-106
		0, 3	l ''	''		0.	0112011	3-thienyl	
20	444	CF ₃	н	н	F	CI	CH ₂ cPr	5-methyl	
		J						3-thienyl	
	445	CF ₃	Н	н	F	Ci	CH ₂ cPr	3-thienyl	113-115
	446	CF ₃	Н	н	F	CI	CH ₂ cPr	Ph-2,4-F ₂	
	447	CF ₃	н	Н	F	CI	CH ₂ cPr	Ph-2-F	n ^{23.8} 1.5262
25	448	CF ₃	H	н	F	CI	CH ₂ cPr	Ph-2-F-3-Me	
	449	CF ₃	Н	н	F	CI	CH ₂ cPr	Ph-2-F-4-OMe	
	450	CF ₃	Н	Н	F	CI	CH ₂ cPr	Ph-2-F-4-OMe	
					_			5-Me	
30	451	CF₃	H	Н	F	CI	CH ₂ cPr	Ph-2-F-5-Me	62-63
	452	CF ₃	H	H	F	CI	CH ₂ cPr	Ph-3,5-Me ₂	
	453	CF ₃	H	Н	F	CI	CH ₂ cPr	Ph-3-Me	96-98
	454 455	CF ₃	H	H	F	CI CI	CH ₂ cPr	Ph-3-Me-4-F Ph-3-Me-4-OMe	82-83 72-73
35	456	CF ₃	Н	Н	F	CI	CH₂cPr CH₂cPr	Ph-4-F	76-77
	457	CF ₃	Н Н	Н	, F	CI	CH ₂ cPr	Ph-4-Me	75-76
	458	CF ₃	н	н	F.	CI	CH ₂ cPr	Ph-4-OMe	68-69
	459	CF ₃	н	н	F	Ci .	CH ₂ cPr	Ph	102-104
	460	CF ₃	н	Н	F	CI	CH ₂ C ≡CH	Ph	
40	461	CF ₃	н	н	F	CI	CH ₂ C ≡CH	Ph-4-OMe	
	462	CF ₃	н	н	F	CI	CH ₂ C ≡CH	Ph-2-F-5-Me	
	463	CF ₃	н	н	F	CI	CH ₂ C ≡Cl	Ph-4-OMe	
	464	CF ₃	н	н	F	CI	iPr	Ph	
45	465	CF ₃	н	Н	F	CI	₁iPr	Ph-4-OMe	
	466	CF ₃	н	н	F	CI	iPr	Ph-2-F-5-Me	
	467	CF ₃	н	н	F	CI	nPr	Ph	
	468	CF ₃	н	Н	F	CI	nPr	Ph-4-OMe	
50	469	CF ₃	H	H	F	CI	nPr	Ph-2-F-5-Me	
	470	CF ₃	н	Н	F	CI	nBu 	Ph-4-OMe	
İ	471	CF ₃	Н	Н	F	CI	tBu	Ph-4-OMe	
	472	CF ₃	Н	н	CF ₃	CI	CH ₂ CH=CH ₂	Ph	
	473	CF ₃	Н	H	CF ₃	CI	CH ₂ CH=CH ₂	Ph-4-OMe	
55	474	CF ₃	Н	Н	CF ₃	CI	CH ₂ CH=CH ₂	Ph-2-F-5-Me	
	475 476	CF ₃	H	H	CF ₃	CI CI	CH₂CH₃	Ph Ph 4 OMe	
į	4/0	CF ₃	F1	''	CF ₃	<u> </u>	CH₂CH₃	Ph-4-OMe	

Table 3. (continued)

	(r¹, r² =H)										
5	No.	X1	X2	X3	X4	X5	R¹	R ²	Physical const. mp (°C) refractive index		
	477	CF ₃	H	Н	CF ₃	CI	CH ₂ CH ₃	Ph-2-F-5-Me			
	478	CF ₃	н	н		CI		1			
	479		l "i	H	CF ₃	CI	CH ₂ CH ₂ CI	Ph Ph 4 OM			
10	480	CF ₃	Н	Н	CF ₃	CI	CH ₂ CH ₂ CI	Ph-4-OMe			
	481	CF ₃ CF ₃	Н	i	CF ₃		CH ₂ CH ₂ CI	Ph-2-F-5-Me			
	482			H	CF ₃	CI	CH₃	Ph Ph 4 CM	•		
	483	CF ₃	H	Н	CF ₃	CI	CH₃	Ph-4-OMe			
15	484	CF ₃	H	Н	CF ₃	CI	CH ₃	Ph-2-F-5-Me			
	485	CF ₃		Н	CF ₃	CI	CH ₂ OMe	Ph-4-OMe			
		CF ₃	Н	Н	CF ₃	CI	CH₂cPr	3-methyl pyrazol-1-yl			
	486	CF ₃	Н	Н	CF ₃	CI	CH ₂ cPr	4-methyl			
20	100	05	l l	١		<u> </u>		pyrazol-1-yl			
	487	CF ₃	H	H	CF ₃	CI	CH₂cPr	pyrazol-1-yl			
	488	CF ₃	Н	Н	CF ₃	CI	CH₂cPr	3-methyl			
	489	CE.		l	<u> </u>		OLL and	2-thienyl			
	409	CF ₃	H	H	CF ₃	CI	CH ₂ cPr	4-methyl 2-thienyl			
25	490	CF ₃	н	Н	CF ₃	CI	CH₂cPr	5-methyl			
	+50	0, 3	'''	''	O1 3	"	0112071	2-thienyl			
	491	CF ₃	н	Н	CF ₃	CI	CH ₂ cPr	2-thienyl			
	492	CF ₃	н	Н	CF ₃	CI	CH ₂ cPr	4-methyl			
30	'	3			J 3	•	0.120.1	3-thienyl			
	493	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	5-methyl			
		Ĭ			J	1	_	3-thienyl			
	494	CF ₃	н	н	CF ₃	Ci	CH₂cPr	3-thienyl			
35	495	CF ₃	н	Н	CF ₃	CI	CH ₂ cPr	Ph-2,4-F ₂			
35	496	CF ₃	н	н	CF ₃	CI	CH₂cPr	Ph-2-F			
	497	CF ₃	Н	Н	CF ₃	CI	CH₂cPr	Ph-2-F-3-Me			
	498	CF ₃	н	Н	CF ₃	CI	CH₂cPr	Ph-2-F-4-OMe			
40	499	CF ₃	Н	н	CF ₃	CI	CH₂cPr	Ph-2-F-4-OMe 5-Me			
-	500	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-2-F-5-Me			
	501	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-3,5-Me ₂			
	502	CF ₃	н	Н	CF ₃	CI	CH ₂ cPr	Ph-3-Me			
	503	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-3-Me-4-F			
45	504	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-3-Me-4-OMe			
	505	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-4-F			
	506	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-4-Me			
	507	CF ₃	н	н	CF ₃	CI	CH ₂ cPr	Ph-4-OMe			
50	508	CF3	' н	н	CF ₃	CI	CH ₂ cPr	Ph	65-67		
-	509	CF ₃	н [н	CF ₃	CI	CH ₂ C≖CH	Ph			
	510	CF ₃	н [н	CF ₃	CI	CH ₂ C≡CH	Ph-4-OMe			
	511	CF ₃	- н	н	CF ₃	CI	CH ₂ C≔CH	Ph-2-F-5-Me	ļ		
	512	CF ₃	н	н	CF ₃	CI	CH ₂ C=CI	Ph			
55	513	CF ₃	- н	н	CF ₃	CI	iPr	Ph	.]		
	514	CF ₃	н	Н	CF ₃	CI	lPr	Ph-4-OMe			
	515	CF ₃	н	н	CF ₃	CI	iPr	Ph-2-F-5-Me	,		
	لــــــــــا	-· <u>3</u>	1	i	3						

Table 3. (continued)

	(r¹, r² =H)										
5	No.	X ¹	X2	Хз	X4	X5	R ¹	R ²	Physical const. mp (°C) refractive index		
	516	CE	H	Н		CI	nPr	Ph	IIIdex		
	517	CF ₃]		CF ₃	1					
	1	CF ₃	H	Н	CF ₃	CI	nPr	Ph-4-OMe			
10	518	CF ₃	H	Н	CF ₃	CI	nPr	Ph-2-F-5-Me			
	519	CF ₃	H	Н	CF ₃	CI	nBu	Ph-4-OMe			
	520	CF ₃	H	H	CF ₃	CI	tBu	Ph-4-OMe			
	521	CF ₃	H	H	CF ₃	F	CH ₂ CH=CH ₂	Ph			
15	522	CF ₃	Н	H	CF ₃	F	CH ₂ CH=CH ₂	Ph-4-OMe			
	523	CF ₃	H	H	CF ₃	F	CH ₂ CH=CH ₂	Ph-2-F-5-Me			
	524	CF ₃	H	Н	CF ₃	F	CH₂CH₃	Ph			
	525	CF ₃	H	H	CF ₃	F	CH ₂ CH ₃	Ph-4-OMe			
	526	CF ₃	H	H	CF ₃	F	CH ₂ CH ₃	Ph-2-F-5-Me			
20	527	CF ₃	H	H	CF ₃	F	CH ₂ CH ₂ CI	Ph Dis 4 OM			
	528 529	CF₃	H	H	CF ₃	F	CH ₂ CH ₂ CI	Ph-4-OMe			
	530	CF₃	H	H	CF ₃	F	CH ₂ CH ₂ CI	Ph-2-F-5-Me			
	530	CF₃	Н	H	CF ₃	F	CH₃	Ph Ph			
25	532	CF ₃	H H	H	CF ₃	F	CH ₃	Ph-4-OMe			
	533	_		H	CF ₃	F	CH ₃	Ph-2-F-5-Me			
	534	CF ₃		Н	CF₃	F	CH ₂ OMe	Ph-4-OMe			
	334	CF ₃	"	"	CF ₃	[CH₂cPr	3-methyl 2-thienyl			
	535	CF ₃	Н	Н	CF3	F	CH ₂ cPr	4-methyl			
30	555	0.3	l ''	''	0,3	'	0112077	2-thienyl			
	536	CF3	Н	Н	CF ₃	F	CH ₂ cPr	5-methyl			
		J			- 3			2-thienyl			
	537	CF ₃	н	н	CF ₃	F	CH ₂ cPr	2-thienyl			
35	538	CF ₃	н	Н	CF ₃	F	CH ₂ cPr	4-methyl			
		_					_	3-thienyl			
	539	CF ₃	н	н	CF ₃	F	CH ₂ cPr	5-methyl			
								3-thienyl			
	540	CF3	н	Н	CF ₃	F	CH ₂ cPr	3-thienyl			
40	541	CF ₃	Н	н	CF ₃	F	CH₂cPr	4-methyl			
								3-thienyl			
	542	CF ₃	Н	Н	CF ₃	F	CH ₂ cPr	5-methyl			
		0.5		,,		_		3-thienyl			
45	543	CF ₃	H	Н	CF ₃	F	CH₂cPr	3-thienyl			
	544	CF ₃	Н	Η :	CF ₃	F	CH₂cPr	Ph-2,4-F ₂			
	545	CF ₃	H	H	CF ₃	F	CH₂cPr	Ph-2-F			
	546	CF₃	Ι:	н	CF ₃	F	CH ₂ cPr	Ph-2-F-3-Me			
	547	CF ₃	H	н	CF ₃	F	CH₂cPr	Ph-2-F-4-OMe			
50	548	CF ₃	н	н	CF ₃	F	CH ₂ cPr	Ph-2-F-4-OMe			
	549	ر ا	ا ں	ایا	Cr	[CH *D*	5-Me			
	550	CF ₃	Н	Н	CF ₃	F	CH ₂ cPr	Ph-2-F-5-Me			
	550 551	CF ₃	H	Н	CF ₃	F	CH ₂ cPr	Ph-3,5-Me ₂			
55	552	CF ₃		Н	CF ₃	F	CH ₂ cPr	Ph-3-Me			
	553	CF ₃	H H	H	CF ₃	F	CH ₂ cPr	Ph-3-Me-4-F			
	554	CF ₃	н	H	CF ₃	F	CH ₂ cPr	Ph-3-Me-4-OMe Ph-4-F			
		CF ₃	''	''	CF ₃	F	CH ₂ cPr	F11-4-F			

Table 3. (continued)

	$(r^1, r^2 = H)$									
5	No.	X ¹	X2	X3	X ⁴	X ⁵	R ¹	R ²	Physical const. mp (°C) refractive index	
	555	CF3	Н	Н	CF ₃	F	CH ₂ cPr	Ph-4-Me		
	556	CF ₃	н	Н	CF ₃	F	CH ₂ cPr	Ph-4-OMe		
10	557	CF ₃	Н	н	CF ₃	F	CH ₂ cPr	Ph		
,,,	558	CF ₃	Н	Н	CF ₃	F	CH₂C≡CH	Ph		
	559	CF ₃	н	Н	CF ₃	F	CH ₂ C≡CH	Ph-4-OMe		
	560	CF ₃	н	Н	CF ₃	F	CH ₂ C≕CH	Ph-2-F-5-Me		
	561	CF ₃	Н	Н	CF ₃	F	CH ₂ C≡CI	Ph-4-OMe		
15	562	CF ₃	Н	н	CF ₃	F	iPr	Ph		
	563	CF ₃	н	н	CF ₃	F	iPr	Ph-4-OMe		
	564	CF ₃	н	н	CF ₃	F	iPr	Ph-2-F-5-Me		
	565	CF3	Н	н	CF ₃	F	nPr	Ph		
20	566	CF ₃	н	н	CF ₃	F	nPr	Ph-4-OMe		
	567	CF ₃	н	н	CF ₃	F	nPr	Ph-2-F-5-Me		
	568	CF ₃	Н	н	CF ₃	F	nBu	Ph-4-OMe		
	569	CF ₃	Н	н	CF ₃	F	tBu	Ph-4-OMe		
	570	CF ₃	F	н	Н	CI	CH ₂ cPr	Ph		
25	571	CF ₃	F	н	н	CI	CH ₂ cPr	Ph-4-OMe		
	572	CF ₃	F	н	н	CI	CH ₂ cPr	Ph-2-F-5-Me		
	573	CF ₃	F	н	Н	F	CH ₂ cPr	Ph		
	574	CF ₃	F	н	Н	F	CH ₂ cPr	Ph-4-OMe		
30	575	CF ₃	F	н	н	F	CH ₂ cPr	Ph-2-F-5-Me		
	576	CF ₃	CF ₃	Н	. н	CI	CH ₂ cPr	Ph		
	577	CF3	CF ₃	Н	Н	CI	CH₂cPr	Ph-4-OMe		
	578	CF ₃	CF ₃	Н	Н	CI	CH ₂ cPr	Ph-2-F-5-Me	1	
	579	CF ₃	CF ₃	Н	Н	F	CH₂cPr	Ph		
35	580	CF ₃	CF ₃	Н	Н	F	CH ₂ cPr	Ph-4-OMe		
	581	CF ₃	CF ₃	н	Н	F	CH ₂ cPr	Ph-2-F-5-Me		
	582	CF ₃	CI	н	Н	CF ₃	CH ₂ cPr	Ph		
	583	CF ₃	CI	н	Н	CF ₃	CH ₂ cPr	Ph-4-OMe		
40	584	CF ₃	CI	Н	Н	CF ₃	CH₂cPr	Ph-2-F-5-Me		
	585	CF ₃	F	Н	н	CF ₃	CH₂cPr	Ph		
	586	CF ₃	F	Н	н	CF ₃	CH ₂ cPr	Ph-4-OMe		
	587	CF ₃	F	Н	Н	CF ₃	CH₂cPr	Ph-2-F-5-Me		
	588	CF ₃	CF ₃	Н	H	CF ₃	CH₂cPr	Ph		
45	589	CF ₃	CF ₃	Н	Н	CF ₃	CH ₂ cPr	Ph-4-OMe		
	590	CF ₃	CF ₃	Н	Н	CF ₃	CH ₂ cPr	Ph-2-F-5-Me		
	591	CF ₃	H	Н	H	CH ₃	CH ₂ cPr	Ph		
	592	CF ₃	н	H	H	CH ₃	CH₂cPr	Ph-4-OMe		
50	593	CF ₃	H	Н	H	CH₃	CH ₂ cPr	Ph-2-F-5-Me		
I	594	CF ₃	Н	Н	F	CH ₃	CH ₂ cPr	Ph		
	595	CF ₃	Н	н	F	CH₃	CH ₂ cPr	Ph-4-OMe		
- 1	596	CF ₃	H	Н	F	CH₃	CH ₂ cPr	Ph-2-F-5-Me		
ĺ	597	CF ₃	H	Н	CI	CH ₃	CH ₂ cPr	Ph		
55	598	CF ₃	Н	H	CI	CH₃	CH ₂ cPr	Ph-4-OMe		
į	599	CF ₃	H	Н	CI	CH ₃	CH ₂ cPr	Ph-2-F-5-Me		
L	600	CF ₃	F	н	F	F	CH ₂ cPr	Ph		

Table 3. (continued)

							(r ¹ , r ² =H)		
5	No.	X ¹	X ²	Хз	X ⁴	X5	R ¹	R ²	Physical const. mp (°C) refractive index
	601	CF ₃	F	Н	F	F	CH ₂ cPr	Ph-4-OMe	
	602	CF ₃	F	Н	F	F	CH ₂ cPr	Ph-2-F-5-Me	
10	603	CF ₃	F	Н	F	CI	CH ₂ cPr	Ph	
10	604	CF ₃	F	Н	F	CI	CH ₂ cPr	Ph-4-OMe	
	605	CF ₃	F	Н	F	CI	CH ₂ cPr	Ph-2-F-5-Me	
	606	CF ₃	F	H	CI	F	CH ₂ cPr	Ph	
	607	CF ₃	F	H	Ci	F	CH ₂ cPr	Ph-4-OMe	
15	608	CF ₃	F	Н	Ci	F	CH ₂ cPr	Ph-2-F-5-Me	
	609	CF ₃	F	H	Ci	CI	CH ₂ cPr	Ph	
	610	CF ₃	F	Н	Ci	CI	CH ₂ cPr	Ph-4-OMe	
	611	CF ₃	F	Н	ci	CI	CH ₂ cPr	Ph-2-F-5-Me	
20	612	CF ₃	CI	Н	F	F	CH ₂ cPr	Ph	
20	613	CF ₃	Ci	н	F	F	CH ₂ cPr	Ph-4-OMe	
	614	CF ₃	Ci	H	F	F	CH ₂ cPr	Ph-2-F-5-Me	
	615	CF ₃	Ci	Н	F	CI	CH ₂ cPr	Ph	
	616	CF ₃	CI	Н	F	CI	CH ₂ cPr	Ph-4-OMe	
25	617	CFa	CI	Н	F	CI	CH ₂ cPr	Ph-2-F-5-Me	
	618	CF ₃	CI	Н	CI	F	CH ₂ cPr	Ph	
	619	CF ₃	CI	н	CI	F	CH ₂ cPr	Ph-4-OMe	
	620	CF ₃	CI	н	CI	F	CH ₂ cPr	Ph-2-F-5-Me	
	621	CF ₃	CI	н	CI	CI	CH ₂ cPr	Ph	
30	622	CF ₃	CI	н	CI	CI	CH ₂ cPr	Ph-4-OMe	
	623	CF ₃	CI	н	CI	Ci	CH ₂ cPr	Ph-2-F-5-Me	
	624	CF ₃	CI	Н	CI	CF ₃	CH ₂ cPr	Ph	
	625	CF ₃	CI	н	F	CF ₃	CH ₂ cPr	Ph	
35	626	CF3	F	н	CI	CF ₃	CH ₂ cPr	Ph	
	627	CF ₃	F	н	F	CF ₃	CH ₂ cPr	Ph	
	628	CF ₃	CF ₃	н	CI	ÇĨ	CH ₂ cPr	Ph	
	629	CF ₃	CF ₃	н	CI	P	CH ₂ cPr	Ph	
	630	CF ₃	CF ₃	н	F	CI	CH ₂ cPr	Ph	
40	631	CF ₃	CF ₃	н	F	F	CH ₂ cPr	Ph	
	632	CF ₃	Cľ	н	CF ₃	CI	CH ₂ cPr	Ph	
	633	CF ₃	F	Н	CF ₃	CI	CH ₂ cPr	Ph	
	634	CF ₃	CI	Н	CF ₃	F	CH ₂ cPr	Ph	
45	635	CF ₃	F	н	CF ₃	F	CH ₂ cPr	Ph	
	636	CF ₃	CF ₃	н	CI	CF ₃	CH ₂ cPr	Ph	
	637	CF ₃	CF ₃	н	F	CF ₃	CH ₂ cPr	Ph	
	638	CF ₃	CF3	н	CF ₃	CF ₃	CH ₂ cPr	Ph	
	639	CF ₃	F	F	F	F	CH ₂ cPr	Ph	
50	640	CF ₃	F	F	F	CI	CH ₂ cPr	Ph	
	641	CF ₃	F	F	CI	F	CH ₂ cPr	Ph	
İ	642	CF ₃	F	F	CI	CI	CH₂cPr	Ph	
	643	CF ₃	CI	н	CI	CF ₃	CH₂cPr	Ph-4-OMe	
55	644	CF ₃	CI	н	F	CF ₃	CH₂cPr	Ph-4-OMe	
	645	CF ₃	F.	н	CI	CF ₃	CH ₂ cPr	Ph-4-OMe	
. [646	CF ₃	F	Н	F	CF ₃	CH ₂ cPr	Ph-4-OMe	

Table 3. (continued)

	(r¹, r² =H)									
5	No.	X ¹	X2	Хз	X ⁴	X5	R1	R2	Physical const. mp (°C) refractive index	
	647	CF ₃	CF ₃	Н	CI	CI	CH ₂ cPr	Ph-4-OMe		
	648	CF ₃	CF ₃	н	CI	F	CH ₂ cPr	Ph-4-OMe		
10	649	CF ₃	CF ₃	;;	F	CI	CH ₂ cPr	Ph-4-OMe		
10	650	CF ₃	CF ₃	н	F	F	CH ₂ cPr	Ph-4-OMe		
	651	CF ₃	CI	''	CF ₃	CI	CH ₂ cPr	Ph-4-OMe		
	652	CF ₃	6. F	;;	CF ₃	CI	CH ₂ cPr	Ph-4-OMe		
	653	CF ₃	CI	;;	CF ₃	F	CH ₂ cPr	Ph-4-OMe		
15	654	CF ₃	6.	Н.	CF ₃	F	CH ₂ cPr	Ph-4-OMe		
	655	CF ₃	CF ₃	H	CI	CF ₃	CH ₂ cPr	Ph-4-OMe		
	656	CF ₃	CF ₃	н	F	CF ₃	CH ₂ cPr	Ph-4-OMe		
	657	CF ₃	CF ₃	н	CF ₃	CF ₃	CH ₂ cPr	Ph-4-OMe		
00	658	CF ₃	F	F	F	F	CH ₂ cPr	Ph-4-OMe		
20	659	CF ₃	F	F	F	CI	CH ₂ cPr	Ph-4-OMe		
	660	CF ₃	F	F	CI	F	CH ₂ cPr	Ph-4-OMe		
	661	CF ₃	F	F	ci	CI	CH ₂ cPr	Ph-4-OMe		
	662	CF ₃	н	н	Н	F	CH ₂ -1-F-cPr	Ph		
25	663	CF ₃	н	Н	Н	F	CH ₂ -2-F-cPr	Ph		
	664	CF ₃	н	н	н	F	CH ₂ -2-F ₂ -cPr	Ph		
	665	CF ₃	н	Н	н	F	CH ₂ -1-F-cPr	Ph-4-OMe		
	666	CF ₃	н	н	н	F	CH ₂ -2-F-cPr	Ph-4-OMe		
30	667	CF3	н	н	н	F	CH ₂ -2-F ₂ -cPr	Ph-4-OMe		
30	668	CF3	н	н	CI	CI	CH ₂ -1-F-cPr	Ph		
	669	CF ₃	н	н	CI	CI	CH ₂ -2-F-cPr	Ph		
	670	CF ₃	н	н	CI	CI	CH ₂ -2-F ₂ -cPr	Ph		
	671	CF ₃	н	н	CI .	CI	CH ₂ -1-F-cPr	Ph-4-OMe		
35	672	CF ₃	н	Н	CI	CI	CH ₂ -2-F-cPr	Ph-4-OMe		
	673	CF ₃	н	н	CI	CI	CH ₂ -2-F ₂ -cPr	Ph-4-OMe		
	674	CF ₃	н	н	CI	F	CH ₂ -1-F-cPr	Ph		
	675	CF ₃	н	н	CI	F	CH ₂ -2-F-cPr	Ph		
40	676	CF ₃	Н	н	CI	F	CH ₂ -2-F ₂ -cPr	Ph		
10	677	CF ₃	Н	н	CI	F	CH ₂ -1-F-cPr	Ph-4-OMe		
	678	CF ₃	н	н -	CI	F	CH ₂ -2-F-cPr	Ph-4-OMe		
	679	CF ₃	Н	н	CI	F	CH ₂ -2-F ₂ -cPr	Ph-4-OMe		
	680	CF ₃	н	н	F	CI	CH ₂ -1-F-cPr	Ph		
45	681	CF ₃	Н	н	F	CI	CH ₂ -2-F-cPr	Ph		
	682	CF ₃	н	н	F	CI	CH ₂ -2-F ₂ -cPr	Ph		
	683	CF ₃	н	н	F	CI	CH ₂ -1-F-cPr	Ph-4-OMe		
	684	CF ₃	Н	Н	F	CI	CH ₂ -2-F-cPr	Ph-4-OMe		
50	685	CF ₃	н	н	F	CI	CH ₂ -2-F ₂ -cPr	Ph-4-OMe		
	686	CF ₃	н	Н	F	F	CH ₂ -1-F-cPr	Ph		
]	687	CF ₃	н	Н	F	F	CH ₂ -2-F-cPr	Ph		
	688	CF ₃	н	н	F	F	CH ₂ -2-F ₂ -cPr	Ph		
ľ	689	CF ₃	н	Н	F	F	CH ₂ -1-F-cPr	Ph-4-OMe		
55	690	CF ₃	н	Н	F	F	CH ₂ -2-F-cPr	Ph-4-OMe		
į	691	CF ₃	Н	H	F	F	CH ₂ -2-F ₂ -cPr	Ph-4-OMe		
į	692	CF ₃	н	Н	Н	Br	CH₂CH₃	Ph		

Table 3. (continued)

							(r ¹ , r ² =H)		
5	No.	X1	X ²	X ₃	X4	X ⁵	R ¹	R²	Physical const. mp (°C) refractive index
	693	<u> </u>	H	Н н	H	Br		Ph-4-OMe	muex
		CF ₃	1		1		CH₂CH₃		23.5
	694	CF ₃	Н	H	H	Br	CH ₂ -cPr	Ph	n ^{23.5} 1.5343 n ^{29.5} 1.5330
10	695	CF ₃	H	H	н	Br	CH ₂ -cPr	Ph-4-OMe	n ^{25.5} 1.5330
	696	CF ₃	H	H	l H	Br	CH ₂ -cPr	Ph-2-F-5-Me	
	697	CF ₃	H	H	F	Br	CH ₂ CH ₃	Ph	
	698	CF ₃	H	H	F	Br	CH ₂ CH ₃	Ph-4-OMe	
15	699	CF ₃	H	H	F	Br	CH ₂ -cPr	Ph	114-115
	700	CF ₃	H	H	F	Br	CH ₂ -cPr	Ph-4-OMe	n ^{23.1} 1.5304
	701	CF ₃	H	H	F	Br	CH ₂ -cPr	Ph-4-F	81-82
	702	CF ₃	H	H	CI	Br	CH ₂ CH ₃	Ph	
	703 704	CF ₃	н	H	CI	Br	CH₂CH₃	Ph-4-OMe	
20		CF ₃	Н	H	CI	Br	CH ₂ -cPr	Ph	
	705	CF₃	H	H	CI	Br	CH ₂ -cPr	Ph-4-OMe	
	706 707	CF ₃	H	н н	CI	Br	CH ₂ -cPr	Ph-2-F-5-Me	
	707	CF ₃	H		Br Br	F	CH ₂ CH ₃	Ph	
25	709	CF ₃	l H	Н	Br	F	CH ₂ CH ₃	Ph-4-OMe	1
	710	CF ₃	Н	н	Br	F	CH ₂ -cPr	Ph Ph-4-OMe	
	711	CF ₃	П	н	Br	F	CH ₂ -cPr		
	712	CF ₃	Н	Н	Br	CI	CH ₂ -cPr CH ₂ CH ₃	Ph-2-F-5-Me	
	713	CF ₃	Н		Br	CI	CH ₂ CH ₃	Ph Ph-4-OMe	
30	714	CF ₃	;;	Н	Br	CI	CH ₂ -cPr	Ph-4-OMe	
	715	CF ₃	Н.	н	Br	CI	CH ₂ -cPr	Ph-4-OMe	
ĺ	716	CF ₃	Н.	;;	Br	CI	CH ₂ -cPr	Ph-2-F-5-Me	
	717	CF ₃	Н	н	F	H	CH ₂ -CH ₃	Ph Ph	
35	718	CF ₃	H	н	F	Н	CH ₂ CH ₃	Ph-4-OMe	
	719	CF ₃	Н	н	, F	Н	CH ₂ -cPr	Ph	91-92
	720	CF ₃	Н.	н	F	Н.	CH ₂ -cPr	Ph-4-OMe	109-110
- 1	721	CF ₃	н	н	, F	Н.	CH ₂ -cPr	Ph-2-F-5-Me	109-110
İ	722	CF ₃	Н	н	CI	: н	CH ₂ CH ₃	Ph	
40	723	CF ₃	Н	;;	F	н	CH ₂ -cPr	Ph-2-F	94-95
İ	724	CF ₃	н	H	CI	н	CH ₂ -cPr	Ph	99-100
	725	CF ₃	H	Н	F	н	CH ₂ -cPr	Ph-4-F	110-111
	726	CF ₃	н	н	CI	н	CH ₂ -cPr	Ph-2-F-5-Me	110-111
45	727	CF ₃	Н	н	н	ï	CH ₂ CH ₃	Ph	İ
	728	CF ₃	н	н	н	i	CH ₂ CH ₃	Ph-4-OMe	
l	729	CF ₃	H	н	н	ì	CH ₂ -cPr	Ph	
	730	CF ₃	н	н	н	i	CH ₂ -cPr	Ph-4-OMe	
-	731	CF ₃	н	н	H	i	CH ₂ -cPr	Ph-2-F-5-Me	
50	732	CF ₃	н	н	CI	CI	CH ₂ -cHxe-3	Ph	106-107
	733	CF ₃	H	н	CI	CI	CH ₂ -cHex	Ph	124-125
		-· J			-,	-,	3,4-Br ₂	['"	127-120
	734	CP3	н	н	CI	CI	CH ₂ -cHex	Ph	106-109
55	735	C ₂ F ₅	н	н	F	F	CH ₂ -cPr	Ph	
-	736	C ₂ F ₅	н	н	F	F	CH ₂ -cPr	Ph-4-OMe	
	737	C ₂ F ₅	н	н	F	F	CH ₂ -cPr	Ph-2-F-5-Me	
L	1	-2.5							

Table 3. (continued)

$(r^1, r^2 = H)$									
5	No.	X ¹	X2	Хз	X4	X5	R1	R ²	Physical const. mp (°C) refractive index
	738	C ₂ F ₅	H	Н	F	F	CH ₂ CH ₂ CI	Ph	
	739	C ₂ F ₅	Н	Н	F	F	CH ₂ CH ₃	Ph	
10	740	C ₂ F ₅	н	Н	F '	F	CH ₂ C≡CH	Ph	
10	741	C ₂ F ₅	Н	Н	CI	CI	CH ₂ -cPr	Ph	
	742	C ₂ F ₅	Н	Н	CI	CI	CH ₂ -cPr	Ph-4-OMe	
	743	C ₂ F ₅	Н	Н	CI	CI	CH ₂ -cPr	Ph-2-F-5-Me	
	744	C ₂ F ₅	''	Н	CI	Ci	CH ₂ CH ₂ CI	Ph	
15	745	C ₂ F ₅	;;	H	Ci	Ci	CH ₂ CH ₃	Ph	
	746	C ₂ F ₅	Н	Н	CI	CI	CH ₂ C≡CH	Ph	
	747	C ₂ F ₅	н н	н	F F	CI		Ph	
	748		''	H	F	CI	CH ₂ -cPr	Ph-4-OMe	
	749	C ₂ F ₅	'' H	H		CI	CH ₂ -cPr	Ph-2-F-5-Me	
20		C ₂ F ₅			F		CH ₂ -cPr		
	750 751	C ₂ F ₅	Н	Н	F	CI	CH ₂ CH ₂ Cl	Ph	
	1 1	C ₂ F ₅	Н	H	F	CI	CH₂CH₃	Ph	
	752 753	C ₂ F ₅	Н	H	F	CI	CH ₂ C≔CH	Ph	
25	754	C ₂ F ₅	H	H	CI	F	CH ₂ -cPr	Ph	
	I I	C ₂ F ₅	Н	H	CI	F	CH ₂ -cPr	Ph-4-OMe	
	755	C ₂ F ₅	H	H	CI	F	CH ₂ -cPr	Ph-2-P-5-Me	
	756	C ₂ F ₅	H	H	CI	F	CH ₂ CH ₂ CI	Ph	
	757	C ₂ F ₅	H	H	CI	F _	CH ₂ CH ₃	Ph	
30	758	C ₂ F ₅	Н	H	CI	F	CH ₂ C≡CH	Ph -	
	759	CCI ₃	H	Н	F	F	CH ₂ -cPr	Ph	
	760	CCI3	Н	Н	F	F	CH ₂ -cPr	Ph-4-OMe	
	761	CCl3	Н	Н	F	F	CH ₂ -cPr	Ph-2-F-5-Me	
	762	CCI3	H	Н	F	F	CH ₂ CH ₂ CI	Ph	
35	763	CCI ₃	Н	н	F	F	CH ₂ CH ₃	Ph	
	764	CCI ₃	Н	H	F	F	CH ₂ C⇔CH	Ph	
	765	CCI3	Н	Н	CI	CI	CH ₂ -cPr	Ph	
	766	CCI3	Н	Н	CI	CI	CH ₂ -cPr	Ph-4-OMe	
10	767	CCI3	Н	Н	CI	CI	CH ₂ -cPr	Ph-2-F-5-Me	
	768	CCI3	н	н	CI	CI	CH ₂ CH ₂ CI	Ph	
	769	CCI3	H.	Н	CI.	CI	CH₂CH₃	Ph	
	770	CCI3	Н	Н	Ci	CI	CH ₂ C≔CH	Ph	
	771	CCI3	Н	н	F	CI	CH ₂ -cPr	Ph	
15	772	CCI3	Н	Н	F	CI	CH ₂ -cPr	Ph-4-OMe	
	773	CCI3	н	н	F	CI	CH ₂ -cPr	Ph-2-F-5-Me	
	774	CCI3	н	Н	F	CI	CH ₂ CH ₂ CI	Ph	
	775	CCI3	н	Н	F	CI	CH₂CH₃	Ph	
50	776	CCI3	н	Н	F	CI	CH ₂ C=CH	Ph	
	777	CCI3	Н	н	CI	F	CH ₂ -cPr	Ph	
	778	CCI ₃	н	н	CI	F	CH ₂ -cPr	Ph-4-OMe	
	779	CCI3	н	н	CI	F	CH ₂ -cPr	Ph-2-F-5-Me	
	780	CCI3	н	н	CI	F	CH ₂ CH ₂ CI	Ph	
55	781	CCI ₃	н	Н	Cl	F	CH ₂ CH ₃	Ph	
	782	CCI3	н	н	CI	F	CH ₂ C≡CH	Ph	
	783	CHF ₂	н	н	F	F	CH ₂ cPr	Ph	

Table 3. (continued)

	$(r^1, r^2 = H)$									
5	No.	X ¹	Χs	Хз	X4	X ⁵	R1	R ²	Physical const. mp (°C) refractive index	
	784	CHF ₂	Н	H	F	F	CH ₂ cPr	Ph-4-OMe		
	785	CHF ₂	н	Н	F	F	CH ₂ cPr	Ph-2-F-5-Me		
40	786	CHF ₂	;;	Н	F	F	CH ₂ CH ₂ CI	Ph		
10	787	CHF2	Н	;;	F	F	CH ₂ CH ₃	Ph		
	788	CHF ₂	Н	Н	F	F	CH ₂ C≡CH	Ph		
	789	CHF ₂	Н] ;;	CI	Cı	CH ₂ cPr	Ph		
	790	CHF ₂	;;	H	CI	Ci	CH ₂ cPr	Ph-4-OMe	:	
15	791	CHF ₂	;;	;;	CI	Ci	CH ₂ cPr	Ph-2-F-5-Me		
	792	CHF ₂	;;	н	CI	Ci	CH ₂ CH ₂ CI	Ph		
	793	CHF	Н	Н	Ci	CI		Ph		
	794	CHF ₂	Н	H	Ci	CI	CH ₂ CH ₃ CH ₂ C≡CH	Ph		
	795	CHF ₂	;;	''	F F	CI	_	Ph		
20	796	CHF ₂		H	F	CI	CH₂cPr	1		
	797	CHF ₂	Н	Н	F	CI	CH₂cPr	Ph-4-OMe Ph-2-F-5-Me		
	798	CHF ₂	H	"	F	CI	CH₂cPr	Ph Ph		
	799	CHF ₂	''	;;	F	CI	CH ₂ CH ₂ CI	Ph		
25	800	CHF ₂	'' H	'' H	F	CI	CH ₂ CH ₃	Ph		
	801	CHF ₂	н	''	CI	F	CH ₂ C≡CH	Ph		
	802	CHF ₂	Н Н	''	CI	F	CH ₂ cPr			
	803	CHF ₂	Н	H	CI	F	CH ₂ cPr	Ph-4-OMe Ph-2-F-5-Me		
	804	CHF ₂	Н	Н	CI	F	CH₂cPr	Pn-2-r-5-Me Ph		
30	805	CHF ₂	Н	Н	CI	1	CH ₂ CH ₂ CI	Ph		
	806	CHF ₂	Н	H	CI	F	CH ₂ CH ₃			
	807	- 1	Н	l		F	CH ₂ C≡CH	Ph Dh		
	808	CF ₃	H	H	Н	F	CH₂CF₃	Ph		
35	809	CF ₃ CF ₃		H	H	F	CH ₂ CI	. Ph		
55	810		Н	H	Н	F	CH ₂ CH ₂ Br	Ph		
	811	CF ₃	H	H	Н	F	CH ₂ OEt	Ph		
	812	CF ₃			Н	1	CH ₂ CH ₂ OMe	Ph		
	813	CF ₃	Н	H	Н	F	CH ₂ cPent	Ph		
40	814	CF₃	Н	H	H	F	CH ₂ cHex	Ph		
		CF₃	Н	Н	H	F	CH ₂ cPr-2,2Cl ₂	Ph		
	815 816	CF ₃	H	Н	Н	F	CH ₂ cPr-2,2 Brz	Ph	n ^{23.2} 1.5469	
1	817	CF₃	Н	H	H	F	CH ₂ SMe	Ph	1 0000	
45		CF ₃	Н	H	Н	F	CH ₂ SOMe	Ph		
70	818	CF ₃	Н	н	Н	F	CH ₂ SO ₂ Me	Ph	116-117	
	819	CF₃	Н	Н	Н	F	CH ₂ CO ₂ Me	Ph	n ^{22.8} 1.5194	
	820	CF ₃	Η:	Н	н	F	CH ₂ CO ₂ Et	Ph	n ^{29.8} 1.5167	
	821	CF ₃	Н	н	Н	F	CH ₂ CH ₂ CO ₂ Me	Ph -:		
50	822	CF ₃	Н	Н	Н	F	CH ₂ NMe ₂	Ph :		
	823	CF ₃	Н	Н	Н	F	CH ₂ NHMe	Ph -:		
	824	CF ₃	Н	Н	Н	F	CH ₂ CH ₂ NMe ₂	Ph	24.5	
	825	CF ₃	Н	Н	Н	F	CH ₂ CONH ₂	Ph	n ^{24.5} 1.5380	
	826	CF ₃	Н	Н	Н	F	CH₂CONHMe	Ph	,,	
55	827	CF ₃	Н	Н	Н	F	CH ₂ CONMe ₂	Ph	n ^{23.3} 1.5302	
	828	CF ₃	н	Н	Н	F	CH ₂ CH=CHCF ₃	Ph		
. [829	CF ₃	Н	Н	н	F	CH ₂ CH=CF ₂	Ph		

Table 3. (continued)

	(r¹, r² =H)											
5	No.	X ¹	X2	Хз	X4	X5	R1	R ²	Physical const. mp (°C) refractive index			
	830	CF ₃	Н	Н	Н	F	CH ₂ C(Br)=CH ₂	Ph				
	831	CF ₃	Н	Н	Н	F	CH ₂ C ≡CCH ₃	Ph				
	832	CF ₃	Н	H	"	F	CH ₂ CH ₂ C≡CH	Ph				
10	833	•	H	H		F	CH ₂ C =CCF ₃	Ph				
	834	CF ₃	H	H	F	F		Ph	į l			
		CF ₃		i .	1		CH ₂ CF ₃					
	835	CF ₃	H	H	F	F	CH ₂ CI	Ph				
15	836	CF ₃	H	H	F	F	CH ₂ CH ₂ Br	Ph				
	837	CF ₃	H	H	F	F	CH ₂ OEt	Ph				
	838	CF ₃	Н	Н	F	F	CH ₂ CH ₂ OMe	Ph				
	839	CF ₃	Н	Н	F	F	CH ₂ cPent	Ph				
	840	CF ₃	H	Н	F	F	CH ₂ cHex	Ph				
20	841	CF3	Н	H	F	F	CH ₂ -2,2-Cl ₂ cPr	Ph				
	842	CF ₃	Н	H	F	F	CH ₂ -2,2-Br ₂ cPr	Ph				
	843	CF ₃	H	Н	F	F	CH₂SMe	Ph				
	844	CF ₃	Н	Н	F	F	CH ₂ SOMe	Ph				
	845	CF ₃	Н	Н	F	F	CH ₂ SO ₂ Me	Ph				
25	846	CF ₃	Н	H	F	F	CH ₂ CO ₂ Me	Ph				
	847	CF ₃	Н	H	F	F	CH ₂ CO ₂ Et	Ph	n ^{24.0} 1.5066			
	848	CF ₃	Н	Н	F	F	CH ₂ CH ₂ CO ₂ Me	Ph				
	849	CF ₃	Н	Н	F	F	CH ₂ NMe ₂	Ph				
30	850	CF ₃	н	н	F	F	CH ₂ NHMe	Ph				
50	851	CF3	н	Н	F	F	CH ₂ CH ₂ NMe ₂	Ph				
	852	CF ₃	н	Н	F	F	CH ₂ CONH ₂	Ph	1			
	853	CF ₃	н	н	F	F	CH ₂ CONHMe	Ph				
	854	CF ₃	Н	Н	F	F	CH ₂ CONMe ₂	Ph				
35	855	CF ₃	Н	Н	F	F	CH2CH=CHCF3	Ph				
	856	CF ₃	н	Н	F	F	CH ₂ CH=CF ₂	Ph				
	857	CF ₃	Н	Н	F	F	CH ₂ C(Br)=CH ₂	Ph				
	858	CF ₃	Н	н	F	F	CH ₂ C≔CCH ₃	Ph				
	859	CF ₃	lн	н	F	F	CH2CH2C≡CH	Ph				
40	860	CF ₃	Н	Н	F	F	CH ₂ C≡CCF ₃	Ph				
	861	CF ₃	Н	н	H	F	CH ₂ cPr	Ph-2-Cl	82-83			
	862	CF ₃	Н	н	н	F	CH₂cPr	Ph-4-Br	89-90			
	863	CF ₃	н	H	н	, F	CH ₂ cPr	Ph-2-Cl-4-F				
45	864	CF ₃	н	Н.	н	F	CH ₂ cPr	Ph-2-Br-4-Me				
•	865	CF ₃	н	н	Н	F	CH ₂ cPr	Ph-4-Et				
	866	CF ₃	Н	;;	н	F	CH ₂ cPr	Ph-4-OEt				
	867	CF ₃	Н	Н	Н	F	CH ₂ cPr	Ph-4-OiPr				
	868	CF ₃	H	н	н	F	CH ₂ cPr	Ph-	65-67			
50	868	OFg	''	''	n	F	O112CF1	4-OCH ₂ CH=CH ₂	05-07			
	869	CF ₃	н	н	н	F	CH₂cPr	Ph-	77-78			
	555	⊖r⁻g	'"	''	17	"	0112071	en- 4-OCH ₂ C≡CH	11-10			
	870	CF ₃	н	н	н	F	CH₂cPr	Ph-3-CF ₃	77-78			
	871	CF ₃	''	Н	н Н	F	CH ₂ cPr	Ph-4-CF ₃	124-125			
55	872	CF ₃	Н	Н	Н	F	CH ₂ cPr	-	124-125			
	873		Н	H	Н	F	_	Ph -4-CF ₂ CF ₂ H	06.07			
	5/3	CF ₃		L'1	17	r .	CH ₂ cPr	Ph-4-OCF ₃	96-97			

Table 3. (continued)

							(r ¹ , r ² =H)		
5	No.	X ¹	X2	Хз	X4	X ⁵	R ¹	R ²	Physical const. mp (°C) refractive index
	874	CF ₃	Н	H	Н	F	CH ₂ cPr	Ph-4-OCF ₂ H	
	875	CF ₃	Н	Н	н	F	CH ₂ cPr	Ph-4-OCF ₂ CF ₃	
10	876	CF ₃	н	Н	н	F	CH ₂ cPr	Ph-4-OCF ₂ CF ₂ H	
10	877	CF ₃	Н	Н	Н	F	CH ₂ cPr	1-Me-2-Pyrrolyl	
	878	CF ₃	Н	Н	Н	F	CH ₂ cPr	1-Me-3-Pyrrolyl	
	879	CF ₃	н	Н	н	F	CH ₂ cPr	2-Pyrrolyl	
	880	CF ₃	Н	Н	н	F	CH ₂ cPr	3-Pyrrolyl	
15	881	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-2-Cl	
	882	CF ₃	Ιн	Н	F	F	CH ₂ cPr	Ph-4-Br	
	883	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-2-Cl-4-F	
	884	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-2-Br-4-CH ₃	
20	885	CF ₃	Н	н	F	F	CH ₂ cPr	Ph-4-Et	
20	886	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-4-OEt	
	887	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-4-OiPr	
	888	CF ₃	Н	н	F	F	CH ₂ cPr	Ph-	
		•	1				•	4-OCH ₂ CH=CH ₂	
25	889	CF ₃	н	н	F	F	CH ₂ cPr	Ph-	
							_	4-OCH ₂ C≡CH	
	890	CF ₃	Н	Н	F	F	CH₂cPr	Ph-3-CF ₃	
	891	CF ₃	Н	Н	F	F	CH ₂ cPr	Ph-4-CF ₃	
30	892	CF ₃	H	H	F	F	CH ₂ cPr	Ph-4-CF ₂ CF ₂ H	
50	893	CF ₃	Н	H	F	F	CH ₂ cPr	Ph-4-OCF ₃	
	894	CF ₃	Н	Н	F	F	CH₂cPr	Ph-4-OCF ₂ H	
	895	CF ₃	Н	н	F	F	CH ₂ cPr	Ph-4-OCF ₂ CF ₃	
	896	CF ₃	н	Н	F	F	CH ₂ cPr	Ph-4-OCF ₂ CF ₂ H	
35	897	CF ₃	Н	H	F	F	CH ₂ cPr	1-Me-2-Pyrrolyl	
	898	CF ₃	Н	H	F	F	CH ₂ cPr	1-Me-3-Pyrrolyl	
	899	CF ₃	Н	Н	F	F	CH ₂ cPr	2-Pyrrolyl	
	900	CF ₃	н	Н	F	F	CH ₂ cPr	3-Pyrrolyl	22.0
40	901	CF ₃	н	Н	F	Br	CH ₂ cPr	Ph	n ^{22.0} 1. 5349
40	902	CF ₃	Н	Н	Н	Et	CH ₂ cPr	Ph	
	903	CF ₃	Н	Н	H	nPr	CH ₂ cPr	Ph	
	904	CF ₃	Н	H	Н	nBu	CH ₂ cPr	Ph	
	905	CF ₃	Н	H	Н	tBu	CH ₂ cPr	Ph	İ
45	906	CF ₃	н	H	H	C ₂ F ₅	CH ₂ cPr	Ph	
	907	CF ₃	Н	H	H	OEt	CH₂cPr	Ph	i
	908	CF ₃	Н	Н	Н	OnPr	CH₂cPr	Ph	
	909	CF ₃	н	H	н	OiPr	CH₂cPr	Ph	
50	910	CF ₃	Н	H	Н	OtBu	CH ₂ cPr	Ph	
	911	CF ₃	H	H	OEt	F	CH ₂ cPr	Ph	j
	912	CF ₃	Н	H	OnPr	F	CH ₂ cPr	Ph	
	913	CF ₃	Н	Н	OiPr	F	CH₂cPr	Ph	
	914	CF₃	Н	Н	OtBu	F	CH ₂ cPr	Ph	23.3
55	915	CF ₃	Н	H	Н	SOMe	CH₂cPr	Ph	n ^{23.3} 1.5450
	916	CF ₃	Н	Н	Н	SO ₂ Me	CH ₂ cPr	Ph	144-146
	917	CF ₃	Н	Ι	Н	SEt	CH₂cPr	Ph	

Table 3. (continued)

		(r¹, r² =H)												
5	No.	X ¹	X5	Х3	X ⁴	X ⁵	R1	R ²	Physical const. mp (°C) refractive index					
	918	CF ₃	Н	Н	Н	S-iPr	CH ₂ cPr	Ph						
	919	CF ₃	Н	Н	SOMe	F	CH ₂ cPr	Ph	}					
10	920	CF ₃	н	Н	SO ₂ Me	F	CH ₂ cPr	Ph						
	921	CF ₃	н	Н	SEt	F	CH ₂ cPr	Ph						
	922	CF ₃	H	Н	S-iPr	F	CH ₂ cPr	Ph						
	923	CF ₃	Н	Н	OCF ₃	F	CH ₂ cPr	Ph-4-OMe						
	924	CF ₃	н	Н	OCF ₃	F	CH ₂ cPr	Ph-2-F						
15	925	CF ₃	Н	Н	OCF ₃	F	CH ₂ cPr	Ph-4-F						
	926	CF ₃	н	Н	OCF ₃	F	CH ₂ cPr	Ph						
	927	CF ₃	н	Н	NO ₂	OCH ₃	CH ₂ cPr	Ph						
	928	CF ₃	Н	н	NHAc	F	CH ₂ cPr	Ph						
20	929	CF ₃	н	Н	NH ₂	F	CH ₂ cPr	Ph						
	930	CF ₃	Н	н	F	F	CH₂COOH	Ph	113-120					
	931	CF ₃	Н	Н	F	F	CH₂cPr	2-imidazolyl 1-methyl-2-						
25	932	CF ₃	Н	н	F	F	CH₂cPr	-imidazolyl 1-methyl-2-	·					
	933	CF ₃	н	н	F	F	CH ₂ cPr	-oxazolyl						
	934	CF ₃	н	н	F	F	CH ₂ cPr	2-isoxazolyl						
								1-methyl-2-						
30	935	CF ₃	Н	H	F	F	CH ₂ cPr	-isoxazolyl						
	936	CF ₃	Н	Н	F	F	CH ₂ cPr	2-pyrimidinyl						
	937	CF ₃	Н	Н	F	F	CH₂cPr	4,6-dimethyl-2						
	938	CF ₃	н	н	F	F	CH₂cPr	-pyrimidinyl						
	939	CF ₃	н	;;	F	F	CH ₂ cPr	2-thiazolyl 3-chloro-2						
35		0, 3	• •	''	•	•	0112071	thiazolyl						
	940	CF ₃	н	н	F	F	CH ₂ cPr	1-pyrazinyl						
	941	CF ₃	н	н	F	F	CH ₂ cPr	3-mrthyl-1						
		ا ٽ				1	22	pyrazinyl						
40	942	CF ₃	н	н	F	F	CH₂cPr	3-pyridazinyl						
	943	CF ₃	н	н	F	F	CH₂cPr	3-methyl-4						
	944	<u></u>	ایا		_	_	011 - 0	pyridazinyl						
	944	CF ₃	H	H	F	F	CH ₂ cPr	2-furyl						
45	945	CF ₃	Н	Н	F	F	CH ₂ cPr	3-bromo-2furyl						
	540	CF ₃	Н	Н	<u> </u>	F	CH ₂ cPr	4-thiazolyl						

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Table 4.

X2 X1 NOR' O R2

		7	7	7	T			_		,
No.	X 1	Χ²	Χ³	Χª	X3	R1	r ₁	Γ2	R²	Physical const. mp. (°C)/ refractive index
947	CF ₃	H	Н	P	P	CH2 cPr	OMe	н	Ph	25. 4-1. 5069
948	CF.	H	н	F	F	CH₂cPr	Me	Н	Ph	25. 5-1. 5106
949	CF ₃	H	H	F	F	CH2 cPr	Et	Н	Ph	25. 4-1. 5003
950	CF:	H	H	P	P	CH2cPr	OMe	CF,	Рь	25. 5-1. 4855
951	CF3	H	Н	P	P	CH ₂ cPr	P	Н	Ph	25. 5-1. 5059
952	CP,	H	H	P	P	СН₂сРг	CI	H	Ph	25. 5-1. 5244
953	CF.	Н	Н	P	P	CH2cPr	SMe	H	Ph	25. 5~1. 5220
954	CF ₃	H	н	F	F	CH2cPr	=0		Pb	79-81
955	CF,	H	H	F	F	CH ₂ cPr	Me	Me	Ph	25. 6-1. 5105
956	CF3	H	H	F	P	CH2 cPr	P	F	Pb	50-52
957	CF3	H	H	h	P	CH2cPr	NHMe	Ħ	Ph	
958	CF ₃	Н	H	Cl	P	CH2cPr	OMe	H	Ph	
959	CF,	H	H	Cl	F	CH₂cPr	Me	H	Ph	
960	CF3	H	Н	CI	F	CH₂cPr	Et	Н	Ph	
961	CF3	H	H	CI	P	CH2cPr	OMe	CF ₃	Ph	
962	CP ₃	Н	Н	Cl	P	CH2cPr	F	Н	Ph	
963	CF ₃	H	Н	Cl	P	CH2cPr	Cl	Н	Ph	
964	CF3	H	H	Cl	b	CH2cPr	SMe	Н	Ph	
965	CF.	Н	Н	CI	þ	CH2cPr	=0		Ph	

Table 4 (continued)

	Table 4 (contitued)										
No.	X1	X,	X3	X*	χ5	R1	rı	Γ2	K _s	Physical const. mp. (°C)/ refractive index	
966	CF:	H	Н	CI	P	CH2cPr	Мe	Me	Ph		
967	CF3	H	Н	Cl	F	CH2cPr	F	F	Ph		
968	CF.	Н	H	CI	F	CH2cPr	NHMe	H	Ph		
969	CF,	H	H	P	CI	CH2cPr	OMe	H	Ph		
970	CF ₃	H	H	F	C1	CH2cPr	Me	Н	Ph		
971	CF ₃	H	H	P	Cl	CH2cPr	Et	Н	Ph		
972	CF 3	H	H	P	Cl	CH2cPr	OMe	CF:	Ph		
973	CF3	H	H	P	Cl	CH₂cPr	P	Н	Ph		
974	CF ₃	H	H	F	Cı	CH2cPr	Cl	Н	Ph		
975	CF.	H	H	F	Cl	CH₂cPr	SMe	Н	Ph		
976	CF ₃	H	H	F	C1	CH2cPr	==(3	Ph		
977	CF ₃	H	H	F	C1	CH2cPr	Me	Me	Ph		
978	CF.	H	H	F	CI	CH2cPr	F	P	Ph		
979	CF ₃	Н	H	P	Cl	CH2cPr	NHMe	Н	Ph	·	
980	CF ₃	H	H	Cl	Cl	CH2:Pr	OMe	H	Ph	·	
981	CF 3	H	H	Cı	Cı	CH2cPr	Me	Н	Ph		
982	CF.	H	Ħ	Cl	CI	CH2cPr	Et	Ħ	Ph		
983	CF3	H	H	Cı	CI	CH2cPr	OMe	CF.	Ph		
984	CP3	H	H	CI	CI	CH₂cPr	P	Н	Ph		
985	CF ₃	H	H	CI	CI	CH2CPr	Cl	H	Ph		
986	CF ₃	Н	H	CI	CI	CH2cPr	SMe	H	Ph		
987	CF3	Н	Н	CI	CI	CH2cPr	=0)	Ph		
988	CF ₃	H	R	CI	CI	CH2cPr	Me	Me	Ph		
989	CF ₃	H	H	CI	CI	CH2cPr	F	P	Ph		
990	CF ₃	H	H	CI	CI	CH2cPr	NHMe	Н	Ph		

Table 4 (continued)

	Constituted												
No.	Χı	Χ²	Хэ	Χ·	X3	R¹	rı	Γ2	R²	Physical const. mp.(°C)/ refractive index			
991	CF ₃	Н	H	H	F	CH2cPr	OMe	Н	Ph				
992	CF ₃	H	Н	H	F	CH2cPr	Me	Ħ	Ph				
993	CF ₃	Ħ	Н	H	F	CH₂cPr	Et	Н	Ph				
994	CF3	Н	H	Н	F	CH ₂ cPr	OMe	CF ₃	Ph				
995	CF ₃	H	H	H	P	CH2cPr	P	H	Ph				
996	CF3	H	H	H	F	CH2cPr	Cl	H	Ph				
997	CF3	H	H	H	F	CH2cPr	SMe	H	Ph				
998	CF;	H	H	H	p	CH2cPr	=()	Ph	·			
999	CF ₃	H	H	H	P	CH2cPr	Me	Me	Ph				
1000	CF ₃	н	H	Н	P	CH2cPr	F	F	Ph				
1001	CF,	Н	H	H	F	CH2cPr	NHMe	Н	Ph				
1002	CF3	H	H	P	P	Et	Me	H	Ph				
1003	CF3	H	H	P.	F	CH2C=CH	Ме	H	Ph				
1004	CF ₃	H	H	P	F	CH2CH2C1	Me	H	Ph				
1005	CF.	Ħ	H	P	P	- Et	Мe	H	Pb-4-OMe				
1006	CF ₃	H	H	F	F	CH2C = CH	Me	H	Ph-4-OMe				
1007	CF ₃	H	H	P	P	CH2CH2C1	Me	H	Ph-4-OMe				
1008	CF:	H	H	P	C1	Et	Me	H	Ph				
1009	CF3	Н	H	F	Cl	CH ₂ C≡CH	Me	H	Ph				
1010	CF3	H	Н	P	CI	CH2CH2C1	Ме	Ħ	Ph				
1011	CF ₃	H	Н	P	Cl	Et	Ме	H	Ph-4-OMe				
1012	CF ₃	H	Н	F	CI	CH2C≡CH	Me	Н	Ph-4-OMe	·			
1013	CF,	Н	Н	P	CI	CH2CH2C1	Me	Н	Ph-4-OMe				

Table 4 (continued)

			T	· · · · · · ·		Ψ				
No.	X,	Χ²	χa	Χ°	ΧS	R1	rı	r ₂	R _s	Physical const. mp. (°C)/ refractive index
1014	CF ₃	Н	H	CI	F	Et	Me	Н	Ph	
1015	CF ₃	Н	H	CI	F	CH ₂ C≡CH	Me	Ħ	Ph	
1016	CP3	H	H	CI	P	CH2CH2C1	Me	H	Pb	
1017	CF ₃	H	H	CI	F	Et	Me	H	Pb-4-OMe	
1018	CF:	H	H	Cl	F	CH2C≡CH	Ме	H	Ph-4-OMe	
1019	CF ₃	Н	H	CI	P	CH2CH2C1	Me	H	Ph-4-OMe	
1020	CF,	Н	H	Ci	C1	Et	Ме	Н	Ph	
1021	CF3	Н	H	Cl	Ci	CH ₂ C≡CH	Me	Н	Ph	
1022	CF,	H	H	CI	Cl	CH2CH2C1	Me	Н	Ph	-
1023	CF ₃	Н	H	CI	Cl	Bt	Me	Н	Ph-4-OMe	
1024	CF,	H	H	CI	Cl	CH₂C = CH	Ме	H	Ph-4-OMe	
1025	CF3	H	H	Cl	Cl	CH2CH2C1	Me	H	Ph-4-OMe	
1026	CF.	H	H	H	F	Et	Me	H	Ph	
1027	CF:	Н	H	H	F	CH ₂ C≡CH	Ме	Н	Ph	
1028	CF3	H	H	H	F	CHaCHaC1	Мe	Н	Ph	
1029	CF:	H	H	Н	F	Bt	Me	H	Ph-4-OMe	
1030	CF.	H	H	H	F	CH2C≡CH	Me	H	Ph-4-OMe	
1031	CF.	H	H	H	F	CH2CH2C1	Ме	H	Ph-4-OMe	
1032	CF3	H	H	P	P	CH2 CN	Н	H	Ph	77-80
1033	CF;	Н	Н	· F	C1	CH2 CN	H	Н	Ph	
1034	CF.	Н	Н	P	F	CH2CH(OEt);	H	Н	Ph	
1035	CF;	H	H	P	Cl	CH2CH(OEt)2	H	H	Ph	
1036	CF,	Н	H	NO ₂	OE t	CH2cPr	Н	Н	Ph	62-63
								_		

*25.4-1.5069 means $n_{D}^{25.4}$ 1.5069 (refractive index) in the above tables.

'H-NMR data (CDCl3. δ ppm from TMS):

- *1 Compound No. 346
- $0.1 \sim 0.2$ (2H. m). $0.45 \sim 0.55$ (2H. m). $0.95 \sim 1.1$ (1H. m), 3.60 (2H. s).
- 3.8~3.9 (2H. m), 3.82 (3H. s), 6.92 (2H. d), 7.1~7.25 (3H. m), 7.51
- 10 (1H. dd). 8.53 (1H. brs)

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[0045] The compounds of the present invention can show excellent fungicidal activity against wide varieties of fungi, and therefore, the compounds can be useful for plant disease control in the farming of agricultural and horticultural crops including ornamental flowers, turf and forage crops.

[0046] The examples for the plant diseases to be controlled with the compounds of the present invention are given in the following.

Paddy rice	Blast	(Pyricularia oryzae)
	Sheath blight	(Rhizoctonia solani)
	Bakanae disease	(Gibberella fujikuroi)
Barley Wheat	Heiminthosporium Loose smut Scab	leaf spot (Cochliobolus miyabeanus) (Ustilago nuda) (Gibberella zeae)
	Leaf rust	(Puccinia recondita)
	Eye spot	(Pseudocercosporella herpotricholdes)
	Glume blotch	(Leptosphaeria nodorum)
	Powdery mildew	(Erysiphe graminis f. sp. tritici)
Potato Ground nut Sugar beat Cucumber	Fusarium snow blight Late blight Leaf spot Cercospora leaf spot Powdery mildew	(Micronectriella nivalis) (Phytophthora infestans) (Mycosphaerella aradius) (Cercospora beticola) (Sphaerotheca fuliginea)
	Sclerotinia rot	(Sclerotinia sclerotiorum)
	Gray mold	(Botrytis cinerea)
Tomato	Downy mildew Leaf mold Late blight	(Pseudoperonospora cubensis) (Cladosporium fulvum) (Phytophthora infestans)
Egg plant Onion	Black rot	(Corynespora melongenae)
Strawberry Apple	Gray-mold neck rot Powdery mildew Powdery mildew Scab	(Botrytis allii) (Sphaerotheca humuli) (Podosphaera leucotricha) (Venturia inaequalis)
Persimmon Peach Grape	Blossom blight Anthracnose Brown rot Powdery mildew	(Monilinia mali) (Gloeosporium kaki) (Monilinia fructicola) (Uncinula necator)
	Downy mildew	(Plasmopara viticola)

(continued)

Pear	Rust	(Gymnosporangium asiaticum)
	Black spot	(Alternaria kikuchiana)
Tea-plant	Leaf spot	(Pestalotia theae)
	Anthracnose	(Colletotrichum theae-sinensis)
Orange	Scab	(Elsinoe fawcetti)
	Blue mold	(Penicillium italicum)
Turf	Sclerotinia snow blight	(Sclerotinia borealis)

[0047] In recent years, it is known that various pathogenic fungi have developed their resistance to benzimidazole fungicides and ergosterol biosynthesis inhibitors and that such fungicides have been insufficient in their fungicidal effectiveness. Therefore, it is required to provide new compounds useful as a fungicide which are effective to the resistant-strain of such pathogenic fungi as well. The compounds of the present invention are the ones which can be a fungicide having excellent fungicidal effectiveness not only to the susceptible-strains of pathogenic fungi but also to the resistant-strains of pathogenic fungi to benzimidazole fungicides and ergosterol biosynthesis inhibitors.

[0048] For the preferable examples of plant diseases to be applied with the compounds of the present invention, powdery mildew on wheat, powdery mildew on cucumber, powdery mildew on strawberries, etc. can be given.

[0049] The compounds of the present invention can be utilized as an antifouling agent for preventing the adhesion of aqueous organisms to structures, such as the bottom of a ship and fishing nets, in water and sea.

[0050] Also, the compounds of the present invention can be contained in paints and fibers and thereby used as an antimicrobial agent for walls, bathtubs, shoes and clothes.

[0051] Furthermore, some of the compounds of the present invention can show insecticidal, acaricidal and herbicidal activities.

[0052] In the practical application of the compounds of the present invention obtained as described above, the compounds can be used in the state as it is without formulation, or, for the use as agricultural plant protection chemicals, the compounds can be applied in forms of general formulations for agricultural plant protection chemicals, such as wettable powders, granules, powders, emulsifiable concentrates, aqueous solutions, suspensions and flowables. For the additives and carriers to be used in the formulations described above, vegetable powders, such as soybean powder and wheat powder, mineral fine powders, such as diatomaceous earth, apatite, gypsum, talc, bentonite, pyrophyllite and clay, and organic and inorganic compounds, such as sodium benzoate, urea and Glauber's salt, can be used, when the compounds are formulated into solid formulations. Whereas, when the compounds are formulated into liquid formulations, petroleum fractions, such as kerosine, xylene and solvent naphtha, cyclohexane, cyclohexanone, dimethyl formamide, dimethyl sulfoxide, alcohols, acetone, trichloro ethylene, methylisobutyl ketone, mineral oils, vegetable oils and water, can be used as the solvent. In these formulations, surface active agents may be added to the formulations in order to make the formulations homogeneous and stable, if appropriate.

[0053] The content of the compound of the present invention as the active principle in the formulations is preferably in a range of from 5 to 70%. The wettable powders, the emulsifiable concentrates and the flowable formulation comprising the compound of the present invention prepared as described above can be applied in a form prepared by diluting the formulations with water to the suspension or the emulsion at a desired concentrations, while the powders and the granules of the said compound can be directly applied to plants without dilution.

[0054] The compounds of the present invention can demonstrate sufficient effectiveness on plant diseases independently, however, it is also possible to use the said compound in admixing with 1 or more of other fungicides, insecticides, acaricides or synergists.

[0055] The followings are the examples for the fungicides, insecticides, acaricides, nematocides and plant growth regulators, those which are usable in admixing with the compounds of the present invention.

Fungicides:

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Copper-based fungicides:

[0056] Basic copper chloride, basic copper sulfate, etc.

Sulphur-based fungicides:

[0057] Thiram, maneb, mancozeb, polycarbamate, propineb, ziram, zineb, etc.

Polyhaloalkylthio fungicides:

[0058] Captan. dichlofluanid, folpet, etc.

5 Organochlorine fungicides:

[0059] Chlorothalonil, fthalide, etc.

Organophosphorous fungicides:

[0060] IBP, EDDP, tolclofos-methyl, pyrazophos. fosetyl-Al, etc.

Benzimidazole fungicides:

15 [0061] Thiophanate-methyl, benomyl, carbendazim, thiabendazole, etc.

Dicarboxyimide fungicides:

[0062] Oxycarboxine, mepronyl, flutolanil, techlofthalam, trichlamide, pencycuron, etc.

Acyl alanine fungicides:

[0063] Metalaxyl, oxadixyl, furalaxyl etc.

25 EBI fungicides:

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[0064] Triadimefon, triadomenol, bitertanol, microbutanil, hexaconazol, propiconazole, triflumizole, procloraz, peflazoate, fenarimol, pyrifenox, trifolin, flusilazole, etaconazole, diclobutrazol, fluotrimazole, flutriafen, penconazole, diniconazole, cyproconazole, imazalil, tridemorph, fenpropimorph, buthiobate, etc.

Antibiotics:

[0065] Polyoxin, blasticidin-S, kasugamycin, validamycin, streptomycin sulfate, etc.

35 Others:

[0066] Propamocarb hydrochloride salt, quintozene, hydroxyisoxazole, metasulfocarb, anilazine, isoprothiolane, probenazole, quinomethionate, dithianone, dinocap, dichlomezine, mepaniprim, ferimzone, fluazinam, pyroquilon, tricyclazole, oxolinic acid, dithianone, iminoctazine acetate salt, cymoxanil, pyrrolenitrine, metasulfocarb, diethofencarb, binapacryl, lecithin, sodium hydrogencarbonate, fenaminosulf, dodine, dimethomorph, fenazine oxide, etc.

Insecticides and Acaricides:

Organophosphorous and carbamate insecticides:

[0067] Fenthion, fenitrothion, diazinon, chlorpyrifos. ESP, vamidothion, fenthoate, dimethoate, formothion, malathon, trichlorfon, thiometon, phosmet, dichlorvos, acephate. EPBP, methyl parathion, oxydimeton methyl, ethion, salithion, cyanophos, isoxathion, pyridafenthion, phosalon, methydathion, sulprofos, chlorfenvinphos, tetrachlorvinphos, dimethylvinphos, propaphos, isofenphos, ethylthiometon, profenofos, pyraclophos, monocrotophos, azinphos methyl, aldicarb, methomyl, thiodicarb, carbofuran, carbosulfan, benfuracarb, furathiocarb, propoxur. BPMC, MTMC, MIPC, carbaryl, pirimicarb, ethiofencarb, fenoxycarb, cartap, thiocyclam, bensultap, etc.

Pyrethroid insecticides:

55 [0068] Permethrin, cypermethrin, deltamethrin, fenvalerate, fenpropathrin, pyrethrin. allethrin, tetramethrin, resmethrin, dimethrin, propathrin, fenothrin, prothrin, fluvalinate, cyfluthrin, cyhalothrin, flucythrinate, ethofenprox, cycloprothrin, tralomethrin, silafluophen, brofenprox, acrinathrin, etc.

Benzoyl urea-based insecticides and others:

[0069] Diflubenzuron, chlorfluazuron, hexaflumuron, triflumuron, tetrabenzuron, flufenoxuron. flucycloxuron, buprofezin, pyriproxyfen, methoprene, benzoepin, diaphenthiuron, imidacloprid, fipronil, nicotine sulfate, rotenone, meta-aldehyde, machine oll, Bacillus thuringiensis, microbial insecticides such as insect-pathogenic viruses. etc.

Nematocides:

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[0070] Fenamiphos, phosthiazate, etc.

Acaricides:

[0071] Chlorbenzilate, phenisobromolate, dicofol, amitraz, BPPS, benzomate, hexythiazox, fenbutatin oxide, polynactin, quinomethionate, CPCBS, tetradifon, avermectin. milbemectin, chlofentezin, cyhexatin, pyridaben, fenpyroxymate. tebufenpyrad, pyrimidifen, fenothiocarb, dienochlor, etc.

Plant Growth Regulators:

[0072] Gibberellines (Gibberelline A₃, Gibberelline A₄, Gibberelline A₇, etc.). IAA, and NAA.

[0073] Now, the examples of the formulations comprising the compound of the present invention are described here-inbelow, however, it should be noted that the type and the rate of the additives shall not be limited to the ones described in the examples and can be replaced by wide range of other additives and/or carriers. In the examples hereinbelow, a part mentioned in each of the formulation examples represents a part by weight.

25 Example 3 : Wettable Powder

[0074]

The compound of the present invention	40 parts
Diatomaceous earth	53 parts
Sulfuric acid ester of higher alcohol	4 parts
Alkyl naphthalene sulfonate	3 parts

[0075] All components are admixed and micronized to fine powder, thereby affording the wettable powder formulation containing the active principle at a concentration of 40%.

Example 4: Emulsifiable Concentrate

40 [0076]

The compound of the present invention	30 parts
Xylene	33 parts
Dimethyl formarnide	30 parts
Polyoxy ethlene alkylally ether	7 parts

[0077] All components are admixed and then dissolved into a solution, thereby affording the emulsifiable concentrate formulation containing the active principle at a concentration of 30%.

50 Example 5 : Dust Formulation

[0078]

The compound of the present invention	10 parts
Talc	89 parts
Polyoxy ethlene alkylally ether	1 parts

[0079] All components are admixed and pulverized to fine dusting powder, thereby affording the dust formulation containing the active principle at a concentration of 10%.

Example 6: Granular Formulation

[0080]

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The compound of the present invention	5 parts
Clay	73 parts
Bentonite	20 parts
Sodium salt of dioctylsulfosuccinate	1 part
Sodium phosphate	1 part
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[0081] All components are admixed, pulverized and then kneaded thoroughly while adding water, and then further granulated and dried, thereby affording the granular formulation containing the active principle at a concentration of 5%.

Example 7: Suspension

[0082]

The compound of the present invention	10 parts
Sodium lignin sulfonate	4 parts
Sodium dodecyl benzene sulfonate	1 part
Xanthane Gum	0.2 part
Water	84.8 parts

[0083] All components are admixed and subjected to wet grinding up to the particle size of less than 1μ , thereby affording the suspension containing the active principle at a concentration of 10%.

[0084] Now, the usefulness of the compounds of the present invention as the active principle of a plant protection chemical for controlling various plant diseases is shown in Test Examples described hereinbelow. The effectiveness of the compounds on plant disease control were assessed basing on the pathological changes in the state of plants provided, namely the degree of disease-induced lesion on leaves, stems and other parts of the plants was visually observed, respectively. The assessment was conducted by giving scores in effectiveness on plant diseases to each test plots as the following.

Score, 5; If no lesion were observed.

Score, 4; If the degree of the lesion observed were approximately 10% of the degree in the untreated plot.

Score, 3; If the degree of the lesion observed were approximately 25% of the degree in the untreated plot.

Score, 2; If the degree of the lesion observed were approximately 50% of the degree in the untreated plot.

Score. 1; If the degree of the lesion observed were approximately 75% of the degree in the untreated plot.

Score. 0; If the degree of the lesion observed were almost same as the degree in the untreated plot.

Test Example 1 : Preventive Control Efficacy Test on Wheat Powdery Mildew

[0085] To young seedlings of wheat (Variety: Chihoku) grown in an unglazed pot, the emulsion at a concentration of 12.5 ppm prepared from the emulsifiable concentrate comprising the compound of the present invention were sprayed throughly. After the spraying, the seedlings were dried under natural condition and then inoculated by means of sprinkling with the conidia of the fungus causing wheat powdery mildew (Erysiphe graminia f. sp. tritici) and placed in a greenhouse maintained at around 20 °C for 7 days in order to complete the infection. Appearance of the lesion on leaves cause by the disease was assessed and compared with the lesion on leaves in the untreated plot, thereby evaluating the preventive efficacy of the compound to the disease. The results are shown in Table 5.

Test Example 2 : Preventive Control Efficacy Test on Cucumber Powdery Mildew

[0086] To young seedlings of cucumber (Variety: Sagami-Hanjiro) grown in an unglazed pot, the emulsion at a concentration of 12.5 ppm prepared from the emulsifiable concentrate comprising the compound of the present invention

were sprayed. After the spraying, the seedlings were dried under natural condition and then inoculated by means of sprinkling with the conidia of the fungus causing cucumber powdery mildew (Sphaerotheca fuliginea) and placed in a temperature-controlled room maintained at around 25°C for 11 days in order to complete the infection. Appearance of powdery mildew-causing lesion on the leaves whereto the compound was sprayed was assessed and compared with the lesion appeared on leaves in the untreated plot, thereby evaluating the preventive efficacy of the compound to the disease. The results are shown in Table 5.

[0087] As shown in Table 5, it is demonstrated that the compounds of the present invention can show superior preventive control efficacy to not only wheat powdery mildew but also cucumber powdery mildew in comparison with other compounds tested.

Test Example 3: Test on Cucumber Powdery Mildew Control with Vapour

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[0088] 10 µl of the emulsion at a concentration of 500 ppm prepared from the emulsifiable concentrate comprising the compound of the present invention were fed dropwise onto round aluminium foils each having a diameter of 1 cm and dried at room temperature under natural condition.

[0089] The aluminium foils were then fixed on the upper side of the leaves of cucumber seedlings (Variety: Sagami-Hanjiro) grown in an unglazed pot. After 24 hours, the cucumber leaves were inoculated by means of sprinkling with the conidia of the fungus causing cucumber powdery mildew (Sphaerotheca fuliginea) and placed in a temperature-controlled room maintained at around 25°C for 11 days in order to complete the infection. Appearance of powdery mildew-causing lesion on the leaves placed with the aluminium foil was assessed and compared with the lesion appeared on the leaves in the untreated plot. The control efficacy with the vapour of the compound to the disease was confirmed in case that disease lesion-free circle having a diameter of more than 2 cm is formed around the aluminium foil fixed on a leaf. The results are shown in Table 5.

[0090] On the other hand, other compounds for the comparison, which are disclosed in Japanese Patent Laid-opened No. Hei 2-6453 (see Table 5), did not show the control efficacy with the vapour to the disease.

[0091] Since the compounds of the present invention have vapour action, it is suggested that the compound of the present invention can show plant disease control efficacy even in the inner space of leaves and fruits of crops whereto spraying of a fungicide in even condition is generally rather difficult.

Table 5.

5	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action
10	1	12.5	5	5	
	3	12.5	5	5	
15	4	12.5	5	5	
	5	12. 5	5	5	
	6	12. 5	5	5	
20	7.	12. 5	4	5	
	8	12. 5	5	5	
25	9	12. 5	5	5	
	10	12.5	4	5	
	11	12.5	5	5	
30 ·	12	12. 5	5	5	
	13	12.5	5	5	
	18	12. 5	2	4	
35	19	12. 5	4	5	
	20	12. 5	4	5	
40	22	12. 5	5	5	
	24	12.5	5	5	good
	25	12. 5	5	5	
45	27	12. 5	5	5	
	28	12.5	4	4	

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Table 5 (continued)

5	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action
10	29	10.5			
	1	12. 5	4	4	
15	30	12. 5	5	4	ļ
	32	12.5	4	4	
	33	12.5	3	5	
20	34	12.5	4	5	good
	35	12.5	5	5	good
	36	12.5	5	4	
25	37	12. 5	4	5	good
	38	12. 5	5	4	
	39	12. 5	5	. 5	good
30	42	12. 5	5	5	good
	43	12. 5	5	5	good
<i>35</i>	44	12.5	5	5	good
	45	12.5	4	4	
	46	12. 5	5	5	
40	47	12. 5	5	5	good
	48	12. 5	5	5	good
	49	12.5	5	5	
45	50	1,2. 5	5	5	good
	51	12.5	5	5	good
	52	12.5	5	5	good
50	53	12. 5	5	5	
	54	12.5	5	5	good
55	55	12.5	5	5	
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Table 5 (continued)

o	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action
5	56	12.5	5	5	good
5	57	12.5	5	5	good
	58	12.5	4	4	
0	59	12.5	4	4	
	60	12. 5	4	5	
5	62	12. 5	4	5	
	64	12.5	4	4	
	65	12.5	5	5	
o	66	12.5	5	5	
	68	12.5	5	5	
5	71	12. 5	5	4	
	74	12.5	5	5	
	82	12. 5	5	5	
o	96	12.5	5	5	boog
	99	12.5	5	5	good
5	105	12. 5	5	5	good
	112	12.5	5	5	good
_	113	12.5	5	5	good
9	114	12. 5	4	5	
	115	12.5	4	5	
5	140	12. 5	3	5	good

Table 5 (continued)

5	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action
10	149	12. 5	5	5	good
	156	12.5	5	5	good
15	157	12. 5	5	5	good
	209	12.5	4	4	
	211	12. 5	4	4	
20	212	12. 5	5	5	
	234	12.5	5	5	good
25	239	12. 5	5	5	good
	240	12.5	5	5	good
	241	12. 5	5	5	good
30	242	12. 5	5	5	good
	255	12. 5	5 .	5	good
35	258	12. 5	5	5	good
	260	12.5	5	5	good
	262	12. 5	5	5	good
40	263	12. 5	5	5	
	264	12. 5	5	5	good
45	266	12. 5	5	5	
	267	12.5	5	5	good
50	268	12.5	5	5	
50	269	12. 5	5	5	good
	270	12. 5	5	5	good
55	271	12.5	5	5	

Table 5 (continued)

5	No	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar . action
10	272	12. 5	5	5	good
	273	12.5	5	5	good
15	274	12.5	5	5	
	275	12.5	5	5	
20	276	12.5	5	4	
	277	12.5	5	5	
	279	12.5	5	5	good
25	280	12.5	5	5	
	286	12. 5	5	5	
30 ·	287	12. 5	5	5	
	288	12.5	5	5	
	289	12.5	5	5	
35	290	12. 5	5	5	good
	293	12. 5	4	5	
40	304	12.5	5	5	
	307	12. 5	5	5	,
45	309	12. 5	5	5	good
	311	12. 5	5	5	good
	312	12. 5	5	5	
50	313	12. 5	5	5	good
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Table 5 (continued)

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10	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action
	315	12.5	5	5	
15	316	12.5	5	5	
	319	12.5	5	5	good
20	320	12.5	5	5	
	321	12.5	5	5	
	322	12.5	5	5	
25	323	12.5	5	5	
	324	12. 5	4	4	
30	326	12.5	5	5	İ
	328	12. 5	5	5	good
35	333	12. 5	5	5	good
35	346	12.5	5	4	
	347	12. 5	4	4	
40	348	12. 5	3	4	
	349	12. 5	4	5	
45	351	12. 5	5	4	good
	352	12. 5	5	4	good
	353	12.5	5	4 .	good

Table 5 (continued)

No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action
364	12. 5	5	5	good
367	12. 5	5	5	good
368	12. 5	5	5	good
370	12. 5	5	5	good
371	12. 5	5	5	good
372	12. 5	5	5	good
373	12. 5	5	5	good
374	12.5	5	5	good
375	12.5	5	5	good
376	12.5	5	5	good
377	12. 5	5	5	good
378	12.5	5	5	
379	12. 5	5	5	good
381	12. 5	5	5	good
383	12.5	5	5	good
386	12. 5	5	5	good
389	12.5	5	5	good
391	12.5	5	5	
392	12. 5	5	5	good

Table 5 (continued)

Vapar action

good

good

good

good

good

good

good

good

good

good

good

good

boog

good

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10	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	
	393	12. 5	5	5	
15	395	12.5	5	5	
	396	12.5	5	5	
	399	12.5	4	5	
20	400	12.5	5	5	
	401	12. 5	5	5	
25	402	12.5	5	5	
	403	12.5	5	5	
	404	12. 5	5	5	
30	405	12. 5	5	5	
	406	12. 5	5	5	
35	407	12. 5	5	5	
	408	12. 5	5	5	
40	409	12. 5	5	5	
40	410	12. 5	5	5	
	413	12. 5	5-	5	
45	414	12. 5	5	5	
	415	12. 5	5	5	
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Table 5 (continued)

5	No.	Concentration of active ingredient (ppm)	Wheat Powdery mildew	Cucumber Powdery mildew	Vapar action	
10	430	12.5	5			
	442	12.5	5	5	good	
15	445	12.5	5	5	good	
	447	12.5	5	5	good	
20	451	12. 5	5	5	good	
20	453	12.5	5	5	good	
	454	12.5	5	5	good	
25	455	12. 5	5	5	good	
	456	12. 5	5	5	good	
<i>30</i>	457	12. 5	5	5	good	
	458	12. 5	5	5	good	
	459	12. 5	5	5	good	
35	508	12. 5	5	5		
	694	12. 5	5	·		
40	861	12. 5	4			
40	868	12. 5	4			
	937	12. 5	4			
45	Reference:					
50	C NHCOCH ₂ ONe					

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Industrial Applicability:

[0092] The compounds of the present invention have excellent fungicidal effectiveness, and therefore, are useful as a fungicide for agricultural and horticultural use.

Claims

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1. Benzamidoxime derivatives represented by the formula (i);

$$X_3 \xrightarrow{X_4} X_2 \xrightarrow{H} L^1 L^2$$

$$CE_3 \xrightarrow{NOR_1} 0$$

$$CE_3 \xrightarrow{NOR_1} 0$$

wherein

 R^1 is unsubstituted or substituted C_1 - C_4 alkyl, unsubstituted or substituted C_2 - C_4 alkenyl or unsubstituted or substituted C_2 - C_4 alkynyl,

R² is phenyl optionally having substituents or heterocycle optionally having substituents,

 X^2 , X^3 , X^4 and X^5 are each independently hydrogen, halogen, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_1 - C_4 alkylthio, C_1 - C_4 alkylsulfinyl, C_1 - C_4 alkylsulfonyl, nitro, amino or C_1 - C_4 alkylcarbonylamino,

and r_1 and r_2 are each independently hydrogen, halogen, C_1 - C_4 alkyl, C_1 - C_4 haloalkyl, C_1 - C_4 alkoxy, C_1 - C_4 alkylthlo or amino, or r_1 and r_2 together may form a carbonyl.

2. Benzamidoxime derivatives represented by the formula (I'):

$$\chi_3$$
 χ_4
 χ_5
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wherein

 R^1 is straight-chain or branched C_1 - C_4 alkyl; a group represented by a general formula, R^3CH_2 , wherein R^3 is a group selected from the group consisting of C_3 - C_8 cycloalkyl, C_1 - C_3 haloalkyl, C_1 - C_3 alkoxy, C_1 - C_3 alkylsulfinyl, C_1 - C_2 alkylsulfinyl, C_1 - C_2 alkylsulfinyl, C_1 - C_3 alkylsulfinyl, C_1 - C_2 alkylsulfinyl

 R^2 is phenyl optionally having one or more substituents selected from the group consisting of halogen, C_1 - C_3 alkyl, C_1 - C_3 alkoxy, C_1 - C_3 haloalkyl and C_1 - C_3 haloalkoxy; or 5- or 6-membered aromatic heterocycle containing 1 to 4 heteroatoms or one or more species selected from a group consisting of nitrogen, oxygen and sulphur, those which are optionally having one or more substituents selected from a group consisting of halogen, C_1 - C_3 alkyl, C_1 - C_3 alkoxy, C_1 - C_3 haloalkyl and C_1 - C_3 haloalkoxy,

 X^2 , X^3 , X^4 and X^5 are each independently hydrogen, halogen, C_1 - C_4 alkyl, C_1 - C_4 alkyled, C_1 - C_4 alkyled finyl, C_1 - C_4 alkyled fonyl, nitro, amino or C_1 - C_4 alkyled fonyl amino, and C_1 are each independently hydrogen, halogen, C_1 - C_4 alkyl, C_1 - C_4 alkyled form carbonyl.

3. A process for preparation of benzamidoxime derivatives represented by the formula (I);

wherein R¹, X², X³, X⁴, X⁵, R², r_1 and r_2 are as described above, which comprises reacting a compound represented by the formula (II)

$$\chi^2 \xrightarrow{\text{CF}_3} \text{NOR}^4 \\ \chi^3 \xrightarrow{\chi^4} \chi^5$$

wherein R1, X3, X4 and X5 are as described above, with a compound represented by the formula (III);

wherein Hal represents halogen, and R2, r1 and r2 are as described above.

4. A fungicide for agricultural and horticultural use, comprising one or more of the benzamidoxime derivatives as defined in claim 1 or 2.

40 Patentansprüche

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1. Benzamidoxim-Derivate, angegeben durch die Formel (I);

$$X^{2} \xrightarrow{\text{CF}_{3}} NOR^{1} \xrightarrow{0} R^{2} \cdots (1)$$

worin

 R^1 unsubstituiertes oder substituiertes C_1 - C_4 -Alkyl, unsubstituiertes oder substituiertes C_2 - C_4 -Alkenyl oder unsubstituiertes oder substituiertes C_2 - C_4 -Alkinyl ist,

R² Phenyl, gegebenenfalls Substituenten aufweisend, oder ein Heterocyclus, gegebenenfalls Substituenten aufweisend, ist,

X², X³, X⁴ und X⁵ jeweils unabhängig für Wasserstoff, Halogen, C₁-C₄-Alkyl, C₁-C₄-Halogenalkyl, C₁-C₄-Alk-oxy, C₁-C₄-Halogenalkoxy, C₁-C₄-Alkylthio, C₁-C₄-Alkylsulfinyl, C₁-C₄-Alkylsulfonyl, Nitro, Amino oder C₁-C₄-Alkylcarbonylamino, stehen,

und r_1 und r_2 jeweils unabhängig für Wasserstoff, Halogen, C_1 - C_4 -Alkyl, C_1 - C_4 -Halogenalkyl, C_1 - C_4 -Alkoxy, C_1 - C_4 -Alkylthio oder Amino stehen, oder r_1 und r_2 zusammen ein Carbonyl bilden können.

2. Benzamidoxim-Derivate, angegeben durch die Formel (I');

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 R^1 geradkettiges oder verzweigtes C_1 - C_4 -Alkyl; eine Gruppe, repräsentiert durch die allgemeine Formel R^3 C H_2 , worin R^3 eine Gruppe ist, die aus der Gruppe gewählt wird, bestehend aus C_3 - C_8 -Cycloalkyl, C_1 - C_3 -Halogenalkyl, C_1 - C_3 -Alkylthio, C_1 - C_3 -Alkylsulfinyl, C_1 - C_3 -Alkylsulfonyl, C_1 - C_3 -Alkylsulfonyl, C_1 - C_3 -Alkylsulfonyl, C_1 - C_3 -Alkylsulfonyl, Cyano, Amino, C_1 - C_3 -Monoalkylamino, C_1 - C_3 -Dialkylamino, Acylamino und Cyano- C_2 - C_4 -alkenyl oder C_2 - C_4 -Alkinyl, ist,

 R^2 Phenyl, das gegebenenfalls ein oder mehrere Substituenten aufweist, gewählt aus der Gruppe, bestehend aus Halogen, C_1 - C_3 -Alkyl, C_1 - C_3 -Alkoxy, C_1 - C_3 -Halogenalkyl und C_1 - C_3 -Halogenalkoxy; oder ein 5- oder 6-gliedriger aromatischer Heterocyclus mit 1 bis 4 Heteroatomen oder einer oder mehreren Spezies, gewählt aus der Gruppe, bestehend aus Stickstoff, Sauerstoff und Schwefel, ist, wobei jene, welche gegebenenfalls ein oder mehrere Substituenten aufweisen, von der Gruppe gewählt sind, die aus Halogen, C_1 - C_3 -Alkyl, C_1 - C_3 -Alkoxy, C_1 - C_3 -Halogenalkyl und C_1 - C_3 -Halogenalkoxy besteht,

X², X³, X⁴ und X⁵ jewells unabhängig für Wasserstoff, Halogen, C₁-C₄-Alkyl, C₁-C₄-Halogenalkyl, C₁-C₄-Alk-oxy, C₁-C₄-Halogenalkoxy, C₁-C₄-Alkylthio, C₁-C₄-Alkylsulfinyl, C₁-C₄-Alkylsulfonyl, Nitro, Amino oder C₁-C₄-Alkylcarbonylamino stehen,

und r_1 und r_2 jeweils unabhängig für Wasserstoff, Halogen, C_1 - C_4 -Alkyl, C_1 - C_4 -Halogenalkyl, C_1 - C_4 -Alkoxy, C_1 - C_4 -Alkylthio oder Amino stehen, oder r_1 und r_2 zusammen Carbonyl bilden können.

3. Verfahren zur Herstellung von Benzamidoxim-Derivaten, angegeben durch die Formel (I);

$$X^{2} \xrightarrow{CF_{3}} NOR^{1} \xrightarrow{0} R^{2} \cdots (1)$$

worin R^1 , X^2 , X^3 , X^4 , X^5 , R^2 , r_1 und r_2 wie oben beschrieben sind, welches die Urnsetzung einer Verbindung der Formel (II)

$$X_3 \xrightarrow{X_4} X_8$$
 NH³ · · · · (11)

worin R1, X3, X4 und X5 wie oben beschrieben sind, mit einer Verbindung der Formel (III);

worin Hal für Halogen steht und R^2 , r_1 und r_2 wie oben beschrieben sind, umfasst.

4. Fungizid zur landwirtschaftlichen oder gartenbaulichen Verwendung, umfassend ein oder mehrere der Benzamidoxim-Derivate, wie sie in Anspruch 1 oder 2 definiert sind.

Revendications

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1. Dérivés du benzamidoxime représentés par la formule (I) :

dans laquelle

- R^1 est un alkyle en C_1 à C_4 non substitué ou substitué, un alcényle en C_2 à C_4 non substitué ou substitué ou un alcynyle en C_2 à C_4 non substitué ou substitué,
- R² est un phényle ayant éventuellement des substituants ou un hétérocycle ayant éventuellement des substituants.
- X^2 , X^3 , X^4 et X^5 sont chacun indépendamment un hydrogène, un halogène, un alkyle en C_1 à C_4 , un halogénoalkyle en C_1 à C_4 , un alkyloxy en C_1 à C_4 , un halogénoalcoxy en C_1 à C_4 , un alkylsulfinyle en C_1 à C_4 , un alkylsulfinyle en C_1 à C_4 , un alkylsulfinyle en C_1 à C_4 , un alkylsulfonyle en C_1 à C_4 , un nitro, un amino ou un alkylcarbonylamino en C_1 à C_4 , et C_4 et C_4 et C_4 en alkylsulfonyle en C_1 à C_4 en halogènoalkyle en C_1 à C_4 en alkylsulfonyle en C_1 à C_4 en alkylthio en C_1 à C_4 ou un amino, ou C_1 et C_2 ensemble peuvent former un carbonyle.
- 2. Dérivés du benzamidoxime représentés par la formule (l') :

dans laquelle

 R^1 est un alkyle en C_1 à C_4 à chaîne linéaire ou ramifié ; un groupe représenté par une formule générale, R^3CH_2 , dans laquelle R^3 est un groupe choisi dans le groupe constitué par un cycloalkyle en C_3 à C_8 , un halogénoalkyle en C_1 à C_3 , un alcoxy en C_1 à C_3 , un alkylthio en C_1 à C_3 , un alkylsulfinyle en C_1 à C_3 , un

alkylsulfonyle en C_1 à C_3 , un alcoxycarbonyle en C_1 à C_3 , un cyano, un amino, un monoalkylamino en C_1 à C_3 , un dialkylamino en C_1 à C_3 , un acylamino et un cyano alcényle en C_2 à C_4 ou alcynyle en C_2 à C_4 , C_3 at un phényle ayant éventuellement un ou plusieurs substituants choisis dans le groupe constitué par un halogène, un alkyle en C_1 à C_3 , un alcoxy en C_1 à C_3 , un halogénoalkyle en C_1 à C_3 et un halogénoalcoxy en C_1 à C_3 ; ou un hétérocycle aromatique à 5 ou 6 membres contenant 1 à 4 hétéroatomes ou une ou plusieurs espèces choisies dans le groupe constitué par l'azote, l'oxygène et le soufre, celles qui ont éventuellement un ou plusieurs substituants choisis dans un groupe constitué par un halogène, un alkyle en C_1 à C_3 , un alcoxy en C_1 à C_3 , un halogénoalkyle en C_1 à C_3 et un halogénoalcoxy en C_1 à C_3 , un alkyle en C_1 à C_2 , un alcoxy en C_2 à C_3 , un alcoxy en C_3 à C_4 , un alcoxy en C_4 à C_4 , un alcoxy en C_5 à C_6 , un alkyle en C_7 à C_8 , un alcoxy en C_8 à C_8 , un alcoxy en C_8 à C_8 , un alcoxy en C_8 à C_8 , un alcoxy en C_8 à C_8 , un alcoxy en C_8 à C_8 , un alcoxy en C_8 à C_8 , un alcoxy en C_8 à C_8 , un alkyle

 X^2 , X^4 et X^3 sont chacun independamment un hydrogene, un halogéne, un alkyle en C_1 à C_4 , un alcoxy en C_1 à C_4 , un halogénoalcoxy en C_1 à C_4 , un alkylsulfonyle en C_1 à C_4 , un nitro, un amino ou un alkylcarbonylamino en C_1 à C_4 , et C_1 et C_2 sont chacun indépendamment un hydrogène, un halogène, un alkyle en C_1 à C_4 , un halogénoalkyle en C_1 à C_4 , un alcoxy en C_1 à C_4 , un alkylthio en C_1 à C_4 ou un amino, ou C_1 et C_2 ensemble peuvent former un carbonyle.

3. Procédé de préparation de dérivés du benzamidoxime représentés par la formule (I) :

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dans laquelle R^1 , X^2 , X^3 , X^4 , X^5 , R^2 , r_1 et r_2 sont tels que décrits ci-dessus, qui consiste à faire réagir un composé représenté par la formule (II)

dans laquelle R1, X3, X4 et X5 sont tels que décrits ci-dessus, avec un composé représenté par la formule (III) :

dans laquelle Hal représente un halogène, et R2, r1 et r2 sont tels que décrits ci-dessus.

 Fongicide pour une utilisation agricole et horticole, comprenant un ou plusieurs des dérivés du benzamidoxime selon la revendication 1 ou 2.